



# EIC 2800 SEARCH REPORT



**STIC Database Tracking Number: 316187**

To: BROOKE PURINTON

Location: JEF-3B15

Art Unit: 2881

Friday, December 04, 2009

Case Serial Number: 10/599555

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Location: EIC2800

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## Search Notes

Attached are the edited search results and the search histories from searching EAST databases and CAS/STN Chemical Abstracts.

The search histories are included at the end of this file.

I recommend that you review at least Page 2 to about Page 20 of this file.

After the first 20 pages: The second half of this file has some abstracts about the existence and properties of the isomers explicitly mentioned in the claims.

An apparently comprehensive list of isomers and their half lives is at the "Nuclear Physics A" journal article at this URL: <http://amdc.in2p3.fr/nubase/Nubase2003.pdf>

I've included some data excerpts in the search results.

If you would like more searching on this case, or if you have questions or comments, please notify me.

DERWENT-ACC-NO: 1994-250295

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TITLE: Vascular implant containing two or more radionuclides with specific half lives for long lasting prevention of restenosis or short lived isotope disintegrating to long lived daughter

INVENTOR: FEHSENFELD P; HEHRLEIN C

PRIORITY-DATA: 1993DE-4315002 (May 6, 1993)

**PATENT-FAMILY:**

PUB-NO	PUB-DATE	LANGUAGE
DE 4315002 C1	August 18, 1994	DE
WO 9426205 A1	November 24, 1994	DE
EP 696906 A1	February 21, 1996	DE
JP 08508436 W	September 10, 1996	JA
EP 696906 B1	April 16, 1997	DE
<u>US 5674177 A</u>	October 7, 1997	EN
JP 2735689 B2	April 2, 1998	JA

ABSTRACTED-PUB-NO: DE 4315002 C1

Vascular implant for inhibiting and/or eliminating vascular constrictions contains at least two radionuclides, one with half life 7 hrs. to 7 days, the other with half life over 100 days.

Alternatively, the implant contains a single isotope of half life 7 hrs. to 7 days that disintegrates directly or indirectly to a daughter nuclide of half life over 100 days.

Pref. nuclides are esp. gamma emitters, esp. for short half lives Co 55, Mn 52, Tc 96, Mo 99 and Ni 57 and for long half lives Fe 55, Zn 65 or Co 57.

USE/ADVANTAGE - These implants (stents) are used to prevent restenosis. The entire region where there is a risk of restenosis is irradiated for the entire period that the stent is in place. Initially, the radiation dose is fairly high, subsequently it is moderate but relatively long lasting. The radionuclides can be generated in the implant after this has been made, i.e. they are not applied as a coating which might become detached.

US 5674177 Abstract Text - ABTX (1):

In a vascular implant for the prevention or elimination of vascular restrictions a tubular body to be inserted into a body vessel includes at least a first nuclide species which has a half life in the range of 7 hours to 7 days and a second nuclide species which has a half life of more than 100 days. Instead of providing a second nuclide species, the first nuclide species may also decay into the second nuclide species thereby providing a high initial radioactivity for a relatively short period and a relatively low radioactivity over a relatively long period.

DERWENT-ACC-NO: 1997-153036  
 COPYRIGHT DERWENT INFORMATION LTD

TITLE: Marking and identification of articles comprises use of radioactive isotopes with different half-life periods, applied in determined ratio to article

INVENTOR: LYAPIDEVSKII V K

PRIORITY-DATA: 1993RU-026033 (May 20, 1993)

**PATENT-FAMILY:**

PUB-NO	PUB-DATE	LANGUAGE
RU 2064697 C1	July 27, 1996	RU

An identifying marker is applied to an article being marked and contains radioactive substances. The radioactivities of the substances are registered and are used for the identification of the article. A mixture of radioactive isotopes with different half-life periods is used and their ratios in the identifying marker are varied, so that the number of decays of each isotope is from 100-1000000 decays per sec.. The summed amplitude distribution is measured and acts as a code of the article at the moment of its registration. The code and the date of registration are fixed in a long-term memory and are used to identify the article at any moment of time taking into account the change of the amplitude distribution because of partial decay of the radioactive isotopes in the identifying marker.

USE - Used for the marking and identifying of articles.

ADVANTAGE - The process gives a reduced complexity of application of radioactive substances.

**(57) Abstract:**

FIELD: radiation monitoring. SUBSTANCE: object is marked with an identification sign containing radioactive substances, and its radioactivity is measured and is used for recognizing of the object. Mixtures of

isotopes with different half-life periods are used. The object is identified according to the pulse-height distribution with due account made for partial decay of radioactive isotopes. EFFECT: facilitated procedure.

From RU 2064697 C1, Page 3:

2. Способ по п. 1, отличающийся тем, что опознавательный знак наносят на объект в виде слоя эпоксидной смолы, содержащей радиоактивные изотопы  $\text{Na}^{22}$ ,  $\text{Co}^{60}$ ,  $\text{Nb}^{96}$ ,  
 50  $\text{Sb}^{125}$ ,  $\text{Cs}^{134}$ ,  $\text{Cs}^{137}$ ,  $\text{Bi}^{207}$ , их смеси и соединения.

US-PAT-NO: 5782742

DOCUMENT-IDENTIFIER: US 5782742 A

\*\*See image for Certificate of Correction\*\*

TITLE: Radiation delivery balloon

DATE-ISSUED: July 21, 1998

----- KWIC -----

Detailed Description Text - DETX (34):

The activity of the material that is produced is a function of the new isotope's half-life and irradiation time. The activity A, produced is determined by the formula . . . .

Detailed Description Text - DETX (35):

in which e is the base of the Napierian or natural logarithm (i.e., 2.718 . . . ), and the elapsed time is t.  $\lambda$  designates the decay constant of the isotope that is formed, and is calculated as the Napierian logarithm of two divided by the half life of the isotope. (It is important that time is measured in the same units in all of these calculations.) It should be noted that after a given time, a saturation point is reached, and no further activity is produced by neutron bombardment, i.e., the material is decaying away as fast as it is being produced. This represents an upper limit for the specific activity of a given isotope that can be produced under these conditions.

Detailed Description Text - DETX (52):

As an alternative to P-32, other radionuclides such as Yttrium-90 (Half-life=64.0 hours; Maximum Beta Energy=2.27 Mev; Average Beta Energy=0.9314 MeV; Betas Per Nuclear Disintegration=1.00; Maximum Range in Tissue=1.11 cm); Gold-198 (Half-life=2.696 Days; Maximum Beta Energy=961 keV; Average Beta Energy=316.3 keV; Betas Per Nuclear Disintegration=1.00; Maximum Range in Tissue 0.42 cm); and Iridium-192 (Half-life=74.2 Days; Maximum Beta Energy = 666 kev; Average Beta Energy=222 keV; Maximum Range in Tissue=0.25 cm) may also be useful.

**US 5782742 Claim:**

**23. A radiation delivery balloon catheter as in claim 21, wherein the radiation delivery source is selected from the group consisting of Pt, P-32, Y-90, Au-198, Ir-192, Mo-99, and combinations thereof.**

FYI: data at <http://amdc.in2p3.fr/nubase/Nubase2003.pdf>

G. Audi et al. / Nuclear Physics A 729 (2003) 3–128

Nuclide	Mass excess (keV)	Excitation energy(keV)		Half-life	$J^\pi$	Ens	Reference	Decay modes and intensities (%)
<sup>192</sup> Ir	-34833.2	1.7		73.827 d	0.013	4 <sup>+</sup>	98	$\beta^+ = 95.13\%$ ; $\varepsilon = 4.87\%$
<sup>192</sup> Ir <sup>m</sup>	-34776.5	1.7	56.720	0.005	1.45 m	0.05	1 <sup>-</sup>	98
<sup>192</sup> Ir <sup>n</sup>	-34665.1	1.7	168.14	0.12	241 y	9	(11 <sup>-</sup> )	98
<sup>198</sup> Au	-29582.1	0.6		2.69517 d	0.00021	2 <sup>-</sup>	02	$\beta^- = 100$
<sup>198</sup> Au <sup>m</sup>	-29269.9	0.6	312.2200	0.0020	124 ns	4	5 <sup>+</sup>	02
<sup>99</sup> Mo	-85965.8	1.9		65.94 h	0.01	1/2 <sup>+</sup>	95	$\beta^- = 100$
<sup>99</sup> Mg <sup>m</sup>	-85868.0	1.9	97.785	0.003	15.5 $\mu$ s	0.2	5/2 <sup>+</sup>	95

US-PAT-NO: 5855546

DOCUMENT-IDENTIFIER: US 5855546 A

TITLE: Perfusion balloon and radioactive wire delivery system

DATE-ISSUED: January 5, 1999

INVENTOR-INFORMATION:

NAME	CITY	STATE	ZIP CODE	COUNTRY
Hastings; Roger N.	Maple Grove	MN	N/A	N/A
Urick; Michael J.	Rogers	MN	N/A	N/A

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## Detailed Description Text - DETX (52):

As previously stated, a preferred source of radiation for all embodiments of the present invention is the radioactive compound Nickel-66. **Nickel-66 decays with a half life of 2.28 days with only low energy beta emissions** and no gamma emission into its daughter element **Copper-66**. **Copper-66 then emits high energy beta radiation with a half life of 5.10 minutes** and decays into the stable element Zinc-66. This **two-step decay** has a particular advantage in use in the catheters of the present invention.

## Detailed Description Text - DETX (53):

**The Nickel-66 acts as a carrier for the high energy copper decay** allowing for time to transport the source to the end user, and also allows for disposal of the device through ordinary means in about 23 days. **A Copper-66 source alone would decay quickly and not be useful without the parent Nickel.** Nickel is low cost and has desirable mechanical properties in its pure form and in alloys, such as a Nickel Titanium alloy.

## Detailed Description Text - DETX (55):

**Another preferred radiation source is Gadolinium-153.** Gadolinium-153 is a composite gamma source which can provide low energy gammas to vessel intima layer while providing higher energy gammas to penetrate calcified plaques and reach the adventitia. . . .

FYI: data at <http://amdc.in2p3.fr/nubase/Nubase2003.pdf>

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Nuclide	Mass excess (keV)	Excitation energy(keV)	Half-life			J <sup>π</sup>	Ens	Reference	Decay modes and intensities (%)
<sup>66</sup> Ni	-66006.3	1.4	54.6	h	0.4	0 <sup>+</sup>	98		$\beta^{-}=100$
<sup>66</sup> Cu	-66258.3	0.7	5.120	m	0.014	1 <sup>+</sup>	98		$\beta^{-}=100$

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Nuclide	Mass excess (keV)	Excitation energy(keV)	Half-life			J <sup>π</sup>	Ens	Reference	Decay modes and intensities (%)
<sup>153</sup> Gd	-72889.8	2.5	240.4	d	1.0	3/2 <sup>-</sup>	98		$\epsilon=100$
<sup>153</sup> Gd <sup>IT</sup>	-72794.6	2.5	95.1737	0.0012	3.5 μs	0.4	(9/2 <sup>+</sup> )	98	IT=100
<sup>153</sup> Gd <sup>IT</sup>	-72718.6	2.5	171.189	0.003	76.0 μs	1.4	(11/2 <sup>-</sup> )	98	IT=100

US-PAT-NO: 5873811  
DOCUMENT-IDENTIFIER: US 5873811 A  
TITLE: Composition containing a radioactive component for treatment of vessel wall  
DATE-ISSUED: February 23, 1999

## INVENTOR-INFORMATION:

NAME	CITY	STATE	ZIP CODE	COUNTRY
Wang; Lixiao	Maple Grove	MN	N/A	N/A
Hastings; Roger N.	Maple Grove	MN	N/A	N/A

## CLAIMS:

What is claimed:

1. A method for inhibiting restenosis at a vessel wall region comprising the steps:  
providing an adhesive including a radioactive material; and  
applying said adhesive to said vessel wall region.
2. A method as recited in claim 1 wherein said radioactive material is admixed with said adhesive.
3. A method as recited in claim 1 wherein said radioactive material is chemically bonded to said adhesive.
4. A method as recited in claim 1 wherein said radioactive material is selected from the group consisting of: **Phosphorus 32, Yttrium 90, Iodine 125, Iridium 192, and mixtures thereof.**

[Note, US 6422989 has wording similar to that in US 6019718.]

US-PAT-NO: 6019718  
 DOCUMENT-IDENTIFIER: US 6019718 A  
 TITLE: Apparatus for intravascular radioactive treatment  
 DATE-ISSUED: February 1, 2000

## INVENTOR-INFORMATION:

NAME	CITY	STATE	ZIP CODE	COUNTRY
Hektner; Thomas R.	Medina	MN	N/A	N/A

----- KWIC -----

## Detailed Description Text - DETX (28):

As previously stated, a preferred source of radiation for all embodiments of the present invention is the radioactive compound Nickel-66. **Nickel-66 decays with a half life of 2.28 days with only low energy beta emissions and no gamma emission into its daughter element Copper-66. Copper-66 then emits high energy beta radiation with a half life of 5.10 minutes and decays into the stable element Zinc-66.** This **two-step decay** has a particular advantage in use in the catheters of the present invention.

## Detailed Description Text - DETX (29):

The Nickel-66 acts as a carrier for the high energy copper decay allowing for time to transport the source to the end user, and also allows for disposal of the device through ordinary means in about 23 days. **A Copper-66 source alone would decay quickly and not be useful without the parent Nickel.** Nickel is low cost and has desirable mechanical properties in its pure form and in alloys, such as a Nickel Titanium alloy.

US-PAT-NO: 6045495  
 DOCUMENT-IDENTIFIER: US 6045495 A  
 TITLE: Apparatus and method to treat a disease process in a luminal structure  
 DATE-ISSUED: April 4, 2000

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The radiation source may be a pellet, a wire, an encapsulated radiation source, or an attached radiation source, such as a paste of Ir-192, I-125, or Pd-103. Alternatively, the 10 radiation source may be a  $\gamma$ -radiation emitting isotope, such as, for example, one of the following:  $^{109}\text{Cd}$ ,  $^{113}\text{Sn}$ ,  $^{125}\text{Te}$ ,  $^{125}\text{I}$ ,  $^{93}\text{MO}$ ,  $^{133}\text{Ba}$ ,  $^{145}\text{Sm}$ ,  $^{147}\text{Eu}$ ,  $^{146}\text{Gd}$ ,  $^{157}\text{Tb}$ ,  $^{254}\text{Es}$ ,  $^{242}\text{Am}$ ,  $^{169}\text{Yb}$ ,  $^{186}\text{Re}$ ,  $^{173}\text{Lu}$ ,  $^{172}\text{Hf}$ ,  $^{177}\text{Lu}$ ,  $^{179}\text{Hf}$ ,  $^{183}\text{Re}$ ,  $^{44}\text{Ti}$ ,  $^{178}\text{Hf}$ ,  $^{57}\text{Co}$ ,  $^{178}\text{Hf}$ ,  $^{57}\text{Co}$ ,  $^{181}\text{Rh}$ ,  $^{75}\text{Se}$ ,  $^{123}\text{Te}$ ,  $^{139}\text{Ce}$ ,  $^{151}\text{Ho}$ ,  $^{235}\text{U}$ ,  $^{101}\text{Rh}$ ,  $^{168}\text{Tm}$ ,  $^{176}\text{Lu}$ ,  $^{127}\text{Xe}$ ,  $^{95}\text{Te}$ ,  $^{177}\text{Lu}$ ,  $^{121}\text{Te}$ ,  $^{210}\text{Bi}$ ,  $^{182}\text{Hf}$ ,  $^{203}\text{Hg}$ ,  $^{176}\text{Lu}$ ,  $^{192}\text{Ir}$ ,  $^{194}\text{I}$ ,  $^{150}\text{Eu}$ ,  $^{175}\text{Hf}$ ,  $^{249}\text{Cf}$ ,  $^{88}\text{Zr}$ ,  $^{75}\text{Se}$ ,  $^{210}\text{Bi}$ ,  $^{182}\text{Hf}$ ,  $^{203}\text{Hg}$ ,  $^{176}\text{Lu}$ ,  $^{178}\text{Hf}$ ,  $^{95}\text{Te}$ ,  $^{121}\text{Te}$ ,  $^{203}\text{Hg}$ ,  $^{192}\text{Ir}$ ,  $^{178}\text{Hf}$ ,  $^{150}\text{Eu}$ ,  $^{249}\text{Cf}$ ,  $^{88}\text{Zr}$ .

FYI: data at <http://amdc.in2p3.fr/nubase/Nubase2003.pdf>

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Nuclide	Mass excess (keV)	Excitation energy(keV)	Half-life	$J^\pi$	E <sub>ns</sub>	Reference	Decay modes and intensities (%)
$^{178}\text{Hf}$	-52444.3	2.1	STABLE	$0^+$	94		IS=27.28 7
$^{178}\text{Hf}^m$	-51296.9	2.1	1147.423	0.005	4.0 s 0.2	8 <sup>-</sup> 94	IT=100
$^{178}\text{Hf}^m$	-49998.6	2.1	2445.69	0.11	31 y 1	16 <sup>+</sup> 94	94KLA E IT=100
$^{178}\text{Hf}^p$	-49870.8	2.2	2573.5	0.5	68 $\mu\text{s}$ 2	(14 <sup>-</sup> ) 94	IT=100

**US-PAT-NO:** 6187037  
**DOCUMENT-IDENTIFIER:** US 6187037 B1

**TITLE:** Metal stent containing radioactivatable isotope and method of making same  
**DATE-ISSUED:** February 13, 2001

**INVENTOR-INFORMATION:**

NAME	CITY	STATE	ZIP CODE	COUNTRY
Satz; Stanley	Surfside	FL	33154	N/A

**ABSTRACT:**

A metal stent for vascular implantation comprising a generally tubular structure whose external surface is adapted to engage the interior vascular surface when implanted, said metal of said stent containing a substantially uniform dispersion of from about 0.05 to about 10.00 percent by weight of one or more naturally occurring or enriched stable isotopes having a half-life of less than two months and that are principally beta particle emitters, so that when activated, said stent emits low to moderate dosages of radiation uniformly to reduce cell proliferation.

**US 6187037 Claims Text - CLTX (10):**

10. The stent as described in claim 1 wherein said radio-activatable isotope results in a radioactive isotope selected from the group consisting of antimony-120, antimony-127, barium-128, barium-131, barium-140, bromine-80m, cadmium-115, cerium-134, cerium-141, cerium-143, cobalt-55, copper-64, copper-67, dysprosium-166, erbium-172, gallium-166, gallium-68, germanium-71, gold-198, gold-199, iodine-124, iodine-125, iodine-131, iridium-194, lanthanum-140, lutetium-172, neodymium-140, nickel-66, niobium-95, osmium-191, palladium-100, phosphorus-32, phosphorus-33, platinum-188, platinum-191, platinum-193m, platinum-195m, platinum-197, praseodymium-143, rhenium-186, rhenium-188, rhodium-99, rhodium-101m, rhodium 103m, rhodium-105, rubidium-82, ruthenium-103, scandium-48, silver-111, strontium-82, tantalum-177, tantalum-183, tellurium-132, tellurium-118, terbium-153, terbium-156, thallium-201, thallium-204, thulium-170, thulium-172, tin-117m, tin-121, titanium-45, tungsten-178, ytterbium-166, ytterbium-169, yttrium-87, yttrium-90, yttrium-91, zinc-72, zirconium-89, and mixtures thereof.

[Note, US 6652441 has wording similar to that in US 6217503.]

**US-PAT-NO:** 6217503  
**DOCUMENT-IDENTIFIER:** US 6217503 B1  
**TITLE:** Apparatus and method to treat a disease process in a luminal structure  
**DATE-ISSUED:** April 17, 2001

From US 6217503 Column 10:

The radiation source may be a pellet, a wire, an encapsulated radiation source, or an attached radiation source, such as a paste of Ir-192, I-125, or Pd-103. Alternatively, the radiation source may be a  $\gamma$ -radiation emitting isotope, such as, for example, one of the following:  $^{109}\text{Cd}$ ,  $^{113}\text{Sn}$ ,  $^{127}\text{Te}$ ,  $^{123}\text{I}$ ,  $^{92}\text{Mo}$ ,  $^{129}\text{Ba}$ ,  $^{145}\text{Sm}$ ,  $^{147}\text{Eu}$ ,  $^{148}\text{Gd}$ ,  $^{157}\text{Tb}$ ,  $^{254}\text{Es}$ ,  $^{242}\text{Am}$ ,  $^{169}\text{Yb}$ ,  $^{185}\text{Re}$ ,  $^{173}\text{Lu}$ ,  $^{172}\text{Hf}$ ,  $^{177}\text{Lu}$ ,  $^{179}\text{Hf}$ ,  $^{183}\text{Re}$ ,  $^{146}\text{Tl}$ ,  $^{178}\text{Hf}$ ,  $^{57}\text{Co}$ ,  $^{178}\text{Hf}$ ,  $^{57}\text{Co}$ ,  $^{121}\text{Rh}$ ,  $^{75}\text{Se}$ ,  $^{122}\text{Te}$ ,  $^{139}\text{Ce}$ ,  $^{166}\text{Ho}$ ,  $^{233}\text{U}$ ,  $^{161}\text{Rh}$ ,  $^{166}\text{Tm}$ ,  $^{176}\text{Lu}$ ,  $^{127}\text{Xe}$ ,  $^{95}\text{Te}$ ,  $^{177}\text{Lu}$ ,  $^{121}\text{Te}$ ,  $^{215}\text{Bi}$ ,  $^{182}\text{Hf}$ ,  $^{203}\text{Hg}$ ,  $^{176}\text{Lu}$ ,  $^{192}\text{Ir}$ ,  $^{184}\text{I}$ ,  $^{150}\text{Eu}$ ,  $^{173}\text{Hf}$ ,  $^{249}\text{Cf}$ ,  $^{88}\text{Zr}$ ,  $^{75}\text{Se}$ ,  $^{210}\text{Bi}$ ,  $^{182}\text{Hf}$ ,  $^{202}\text{Hg}$ ,  $^{176}\text{Lu}$ ,  $^{178}\text{Hf}$ ,  $^{85}\text{Te}$ ,  $^{121}\text{Te}$ ,  $^{203}\text{Hg}$ ,  $^{192}\text{Ir}$ ,  $^{178}\text{Hf}$ ,  $^{150}\text{Eu}$ ,  $^{249}\text{Cf}$ ,  $^{88}\text{Zr}$ .

From US 6217503 Column 17:

The radioisotopes that decay with emission of beta plus or beta minus radiation, that have a half-life of between approximately 1 and 72 hours, that have an average decay energy of approximately 500-2000 keV, and that have radiation intensity of greater than or equal to approximately 50%, said radiation intensity being measured in % per decay, may be selected from the group consisting of NA-24, SI-31, K-42, SC-43, SC-44, CO-55, MN-56, CU-61, NI-65, GA-66, GA-68, ZN-71, GA-72, AS-72, SE-73, BR-75, AS-76, BR-76, GE-77, KR-77, AS-78, Y-85, KR-87, ZR-87, NB-89, Y-90, NB-90, SR-91, Y-92, Y-93, ZR-97, IN-111, AG-112, AG-113, SB-122, SN-127, TE-129, BA-139, LA-140, LA-141, LA-142, PR-142, PR-145, TB-148, PM-150, EU-152, HO-166, RE-188, RE-190, IR-194, BI-212, and radioactive sodium-chloride.

From US 6217503 Column 34, Line 48:

Further, isotopes that decay to those listed in the application may be used.

See also US 6217503 Tables 1-4.

FYI: data at <http://amdc.in2p3.fr/nubase/Nubase2003.pdf>

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Nuclide	Mass excess (keV)	Excitation energy(keV)	Half-life	J <sup>π</sup>	E <sub>ns</sub>	Reference	Decay modes and intensities (%)
$^{178}\text{Hf}$	-52444.3	2.1	STABLE	0 <sup>+</sup>	94		IS=27.28 7
$^{178}\text{Hf}^{\star\star}$	-51296.9	2.1 1147.423	0.005	4.0 s	0.2	8 <sup>-</sup>	94 IT=100
$^{178}\text{Hf}^{\star\star}$	-49998.6	2.1 2445.69	0.11	31 y	1	16 <sup>+</sup>	94 94K1A E IT=100
$^{178}\text{Hf}^{\star\star}$	-49870.8	2.2 2573.5	0.5	68 $\mu$ s	2	(14 <sup>-</sup> )	94 IT=100

[Note, US 6287249, US 6458069, and US 6491619 have wording similar to that in US 6261320.]

US-PAT-NO: 6261320

DOCUMENT-IDENTIFIER: US 6261320 B1

\*\*See image for Certificate of Correction\*\*

TITLE: Radioactive vascular liner

DATE-ISSUED: July 17, 2001

INVENTOR-INFORMATION:

NAME	CITY	STATE	ZIP CODE	COUNTRY
Tam; Lisa A.	Lake Forest	CA	N/A	N/A
Trauthen; Brett A.	Newport Beach	CA	N/A	N/A

From Column 27, Line 12:

One preferred embodiment of radioactive-coated stent of the present invention is that which has an isotope layer comprising the gamma-emitting isotope  $^{125}\text{I}$ .

Detailed Description Text - DETX (128):

Yet another preferred embodiment of radioactive coating of the present invention is that which has an isotope layer comprising tungsten-188 (W-188 or  $\sup{188}\text{W}$ ) Tungsten-188 undergoes beta decay to become rhenium-188 (Re-188 or  $\sup{188}\text{Re}$ ). Rhenium-188 undergoes beta decay as well, but emits a much higher energy particle than in W-188 decay.

**The W-188 has a much longer half-life than does Re-188, thus the W-188 almost continuously creates more Re-188.** This process is known as "generator," and the generator isotopes are referred to together by the shorthand W/Re-188 to indicate the relationship between the species. **Generators are attractive for use in radiation delivery devices because they combine the energy levels of a short half-life species with the durability of the long half-life species.** It is a general rule that particle energy and half-life are inversely proportional, and that long half-life species are more economical and practical to work with than short half-life species.

Detailed Description Text - DETX (132):

Combinations of various isotopes provide another preferred embodiment in that, for example, beta-emitting isotopes may be combined with gamma-emitting isotopes where gamma isotopes can deliver dosage to greater depths.

US-PAT-NO: **6416457**  
 DOCUMENT-IDENTIFIER: **US 6416457 B1**

TITLE: System and method for intravascular ionizing tandem radiation therapy  
 DATE-ISSUED: **July 9, 2002**

## INVENTOR-INFORMATION:

NAME	CITY	STATE	ZIP CODE	COUNTRY
Urick; Michael J.	Rogers	MN	N/A	N/A
Verin; Vitali E.	Geneva	N/A	N/A	CH
Popowski; Youri G.	Geneva	N/A	N/A	CH

## Detailed Description Text - DETX (6):

Refer now to FIG. 3 which illustrates the radiation source wire 16 utilized in the system 10 illustrated in FIG. 1. Elongate source wire 16 includes two primary components, namely an elongate shaft 30 and a distally disposed radioactive source 32. Radioactive source 32 may approximate a line source as illustrated. **The radiation source 32 includes a radioisotope emitting ionizing radiation such as beta or gamma radiation.** Preferably, the radioactive source 32 comprises a radiation emitting isotope **such as Sr/Y-90, P-32, Y-90, Ce/Pr-144, Ru/Rh-106, W/Re-188, Ir-192, I-125, or Pd-103.** Radiopaque markers 34 and 36 may be disposed on either side of the radioactive source 32 to facilitate intravascular placement utilizing x-ray fluoroscopy. Elongate source wire 16 may comprise a wide variety of different designs incorporating an elongate shaft 30 and a distally disposed radioactive source 32. Preferably, the source wire 16 comprises the design disclosed in U.S. Pat. No. 5,728,042 to Schwager, which is hereby incorporated by reference.

**US-PAT-NO:** 6685618  
**DOCUMENT-IDENTIFIER:** US 6685618 B2  
**TITLE:** Method for delivering radiation to an intraluminal site in the body  
**DATE-ISSUED:** February 3, 2004

**INVENTOR-INFORMATION:**

NAME	CITY	STATE	ZIP CODE	COUNTRY
Tam; Lisa A.	Lake Forest	CA	N/A	N/A
Trauthen; Brett A.	Newport Beach	CA	N/A	N/A Detailed

----- KWIC -----

**Description Text - DETX (62):**

Yet another preferred embodiment of thin film source of the present invention is that which has an isotope layer comprising tungsten-188 (W-188 or <sup>.sup.188</sup> W). Tungsten-188 undergoes beta decay to become rhenium-188 (Re-188 or <sup>.sup.188</sup> Re). Rhenium-188 undergoes beta decay as well, but emits a much higher energy particle than in W-188 decay. The W-188 has a much longer half-life than does Re-188, thus the W-188 almost continuously creates more Re-188. This process is known as "generator," and these generator isotopes are referred to together by the shorthand W/Re-188 to indicate the relationship between the species. Generators are attractive for use in radiation delivery devices because they combine the energy levels of a short half-life species with the durability of the long half-life species. It is a general rule that particle energy and half-life are inversely proportional, and that long half-life species are more economical and practical to work with than short half-life species.

**US 6685618 Claims 15 and 16:**

14. A method of treating a site within a vessel as in claim 11, wherein said isotope is a gamma or beta emitting isotope.

15. A method as in claim 11, wherein the radioactive isotope comprises an isotope selected from the group consisting of P-32, I-125, Pd-103, W/Re-188, As-73, and Gd-153.

FYI: data at <http://amdc.in2p3.fr/nubase/Nubase2003.pdf>

G. Audi et al. / Nuclear Physics A 729 (2003) 3–128

Nuclide	Mass excess (keV)	Excitation energy(keV)		Half-life		$\beta^+$	Ens	Reference	Decay modes and intensities (%)
<sup>188</sup> Gd	-72889.8	2.5		240.4	d	1.0	$3/2^-$	98	$\varepsilon=100$
<sup>188</sup> Gd <sup>**</sup>	-72794.6	2.5	93.1737	0.0012	3.5 $\mu$ s	0.4	$(9/2^+)$	98	IT=100
<sup>188</sup> Gd <sup>**</sup>	-72718.6	2.5	171.189	0.005	76.0 $\mu$ s	1.4	$(11/2^+)$	98	IT=100
<sup>103</sup> Pd	-87479.1	2.9		16.991	d	0.019	$5/2^+$	01	$\varepsilon=100$
<sup>103</sup> Pd <sup>**</sup>	-86694.3	2.9	784.79	0.10	25 ns	2	$11/2^+$	01	IT=100
<sup>188</sup> W	-38667	3		69.78	d	0.05	$0^+$	02	$\beta^+=100$
<sup>188</sup> Re	-39016.1	1.4		17.0040	h	0.0022	$1^-$	02	$\beta^-=100$

PGPUB-DOCUMENT-NUMBER: 20070272862

DOCUMENT-IDENTIFIER: US 20070272862 A1

TITLE: Method and Device for Remotely Communicating Using Photoluminescence or Thermoluminescence

PUBLICATION-DATE: November 29, 2007

INVENTOR-INFORMATION: Desbrandes, Robert;

Van Gent, Daniel Lee

APPL-NO: 11/569357

DATE FILED: May 23, 2005

## FOREIGN-APPL-PRIORITY-DATA:

COUNTRY	APPL-NO	DOC-ID	APPL-DATE
FR	0405717	2004FR-0405717	May 26, 2004
FR	0503659	2005FR-0503659	April 12, 2005

## PCT-DATA:

APPL-NO: PCT/EP05/52348

DATE-FILED: May 23, 2005

371-DATE: Nov 18, 2006

Brief Summary Text - BSTX (33):

[0047] Each group of entangled photons is made up of emitted photons together or at very short intervals by the same particle of the source, for example: electron, nucleus, atom, molecule.

The sources of ad hoc entangled photons usable for this invention are, for example:

[0048] Natural or artificial radioactive materials producing a radiation in a cascade.

For example, the Cobalt 60 atom emits almost simultaneously two gamma which are entangled and which can be used to irradiate a photoluminescent or thermoluminescent material. . . .

FYI: data at <http://amdc.in2p3.fr/nubase/Nubase2003.pdf>

G. Audi et al. / Nuclear Physics A 729 (2003) 3–128

Nuclide	Mass excess (keV)	Excitation energy(keV)	Half-life	$\beta^+$	E <sub>β</sub>	Reference	Decay modes and intensities (%)
<sup>60</sup> Co	-61649.0	0.6	5.2713 y	0.0008	5 <sup>+</sup>	00	$\beta^- = 100$
<sup>60</sup> Co"	-61590.4	0.6 58.59 0.01	10.467 m	0.006	2 <sup>+</sup>	00	IT≈100; $\beta^- = 0.24$ 3

FYI "Production of Cobalt-60", HW-64073, W.L. Bunch, Hanford Atomic Products Operation, 2-29-1960, Page 2  
<http://www.osti.gov/bridge/servlets/purl/10152247-upHZQJ/native/10152247.pdf>

Cobalt-60 is produced from cobalt-59 in isomeric states. The short half-life (10.4 minutes) isomer decays primarily by gamma emission to the long half-life (5.28 year) isomer which subsequently beta decays to an excited state of nickel 60. Photons of energy 1.17 and 1.33 Mev are emitted to stabilize the nickel 60 nucleus. For long irradiations the existence of the short half-life isomer can be neglected and the differential equation describing the formation of cobalt-60 is:

$$\frac{dN_{60}}{dt} = \phi \sigma_{59} N_{59} e^{-\rho \sigma_{59} t} - \rho \sigma_{60} N_{60} - \lambda^{60} N_{60}$$

where N = number of atoms of cobalt (the subscripts refer to the isotopes),

$\phi$  = conventional neutron flux, n/cm<sup>2</sup> sec,

$\sigma$  = absorption cross section for 2200 m/sec neutrons,

$\lambda$  = decay constant (= 0.693/T) in sec<sup>-1</sup>,

t = exposure time in sec,  
and the superscript refers to the original number of atoms present.

## Accelerated Emission of Gamma Rays from the 31-yr Isomer of $^{178}\text{Hf}$ Induced by X-Ray Irradiation

C. B. Collins, F. Davanloo, M. C. Iosif, R. Dussart,\* and J. M. Hicks  
*Center for Quantum Electronics, University of Texas at Dallas, Richardson, Texas 75083*

From Page 697:

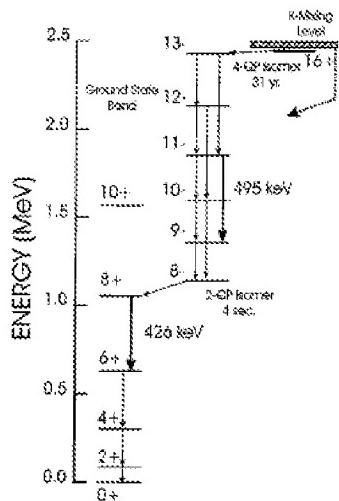


FIG. 3. Energy level diagram for  $^{178}\text{Hf}$  showing a selection of levels relevant to this experiment. The spontaneous decay of the 31-yr,  $16^+$  isomer is shown by the thin arrows. The thick solid arrows show components studied in this work from the particular cascade from the decay forced by the x-ray irradiation. Heavy dotted arrows are inferred transitions needed to feed those shown.

During the spontaneous decay of the  $^{178}\text{Hf}$  isomer the latter, being a member of the GSB, is fed only by cascade from the 4 sec,  $8^-$  bandhead.

FYI: data at <http://amdc.in2p3.fr/nubase/Nubase2003.pdf>

G. Audi *et al.* / Nuclear Physics A 729 (2003) 3–128

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Nuclide	Mass excess (keV)	Excitation energy(keV)	Half-life	$J^\pi$	Ens	Reference	Decay modes and intensities (%)
$^{178}\text{Hf}$	-52444.3	2.1		$0^+$	94		IS=27.28 7
$^{178}\text{Hf}^*$	-51296.9	2.1	1147.423	0.005	4.0	s	IT=100
$^{178}\text{Hf}^*$	-49998.6	2.1	2445.69	0.11	31	y	1
$^{178}\text{Hf}^*$	-49870.8	2.2	2573.5	0.5	68	$\mu\text{s}$	2
							(14 $^-$ ) 94

L128 ANSWER 6 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1997:616665 HCAPLUS

DN 127:337716

OREF 127:66223a,66226a

ED Entered STN: 27 Sep 1997

TI **Half-lives** of isomeric levels of 87mSr, **111mCd**, **113mIn**

and 115mIn activated by 60Co  $\gamma$ -ray irradiation

SO Nuclear Instruments & Methods in Physics Research, Section A: Accelerators, Spectrometers, Detectors, and Associated Equipment (1997), 397(2,3), 478-482

CODEN: NIMAER; ISSN: 0168-9002

PB Elsevier

Journal Contents:

[http://www.sciencedirect.com/science?\\_ob=PublicationURL&\\_tokey=%23TOC%235314%231997%23996029997%2310802%23FLP%23&\\_cdi=5314&\\_pubType=J&\\_auth=y&\\_acct=C000055109&\\_version=1&\\_urlVersion=0&\\_userid=2502287&md5=ec893abac7bdb1577659de47a9e3985c](http://www.sciencedirect.com/science?_ob=PublicationURL&_tokey=%23TOC%235314%231997%23996029997%2310802%23FLP%23&_cdi=5314&_pubType=J&_auth=y&_acct=C000055109&_version=1&_urlVersion=0&_userid=2502287&md5=ec893abac7bdb1577659de47a9e3985c)

AB **Isomers having half-lives of a few hours were photoactivated by 60Co  $\gamma$ -ray irradiation**  
 The **gamma** -rays emitted from the title isomers were measured with a low-background Ge detector. The **decay** data were analyzed by both a least-squares method and maximum likelihood method and the suitability of the latter method is discussed. The obtained **half-lives** are: 87mSr  $2.811 \pm 0.027$  h, **111mCd**  $48.30 \pm 0.15$  min, 113mIn  $1.663 \pm 0.011$  h and **115mIn**  $4.483 \pm 0.005$  h.

IT **Gamma ray**  
 (irradiation; **half-lives** of isomeric levels of 87mSr,  
 111mCd, 113mIn and 115mIn activated by 60Co  $\gamma$ -ray irradiation)

IT Nuclear energy level  
 (isomer; **half-lives** of isomeric levels of 87mSr,  
 111mCd, 113mIn and 115mIn activated by 60Co  $\gamma$ -ray irradiation)

IT **14191-71-0**, Indium 115, properties **14336-64-2**,  
 Cadmium 111, properties  
 RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)  
 (**half-lives** of isomeric levels of 87mSr, 111mCd,  
 113mIn and 115mIn activated by **60Co**  $\gamma$ -ray irradiation)

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

115In

RN 14336-64-2 HCAPLUS

CN Cadmium, isotope of mass 111 (CA INDEX NAME)

111Cd

**L128 ANSWER** 15 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1987:109090 HCAPLUS

DN 106:109090

OREF 106:17751a,17754a

ED Entered STN: 05 Apr 1987

TI **Half-lives** of cadmium-111m, indium-113m, and indium-115m

AU Nemeth, Z.; Lakosi, L.; Pavlicsek, I.; Veres, A.

CS Inst. Isot., Hung. Acad. Sci., Budapest, H-1525, Hung.

SO Applied Radiation and Isotopes (1987), 38(1), 63-4

CODEN: ARISEF; ISSN: 0883-2889

DT Journal

LA English

CC 70-1 (Nuclear Phenomena)

AB Bremsstrahlung from 4.5-MeV e on a Pt target was used to produce nuclear isomers in 3 nuclides through ( $\gamma, \gamma'$ ) **excitation** of higher lying states and their **half-lives** were measured. The obtained values are  $48.54 \pm 0.05$  min for **111mCd**,  $1.660 \pm 0.005$  h for **113mIn**, and  $4.485 \pm 0.004$  h for **115mIn**. The intensity ratio of 497/336 keV  $\gamma$  transitions in the **decay** of **115mIn** was  $1.04 \pm 0.05 + 10^{-3}$ . The methods of anal. and the results are discussed.

IT **Gamma ray**

(scattering of, by cadmium-111 and indium isotopes, inelastic)

IT **14191-71-0**, Indium-115, properties **14336-64-2**, Cadmium-111, properties

RL: PRP (Properties)

(nuclear energy levels of, **excitation** of isomeric, by

$\gamma$ -ray scattering)

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

**115In**

RN 14336-64-2 HCAPLUS

CN Cadmium, isotope of mass 111 (CA INDEX NAME)

**111Cd**

L128 ANSWER 7 OF 42 HCPLUS COPYRIGHT ACS on STN

AN 1997:499756 HCPLUS

DN 127:199354

OREF 127:38487a,38490a

ED Entered STN: 07 Aug 1997

TI Determination of the **elemental distribution in a sample** using neutron-induced gamma-ray emission tomography

AU Spyrou, N. M.; Sharaf, J. M.; Rajeswaran, S.; Mesbahi, E.

CS Dep. Phys., Univ. Surrey, Guildford, GU2 5XH, UK

SO Journal of Radioanalytical and Nuclear Chemistry (1997), 217(2), 237-241

CODEN: JRNCMD; ISSN: 0236-5731

PB Elsevier

DT Journal

LA English

Journal Contents:

<http://springerlink.metapress.com/content/xlrp7g350421/?p=38c9a261e0da4b58903406eb28bb3651&pi=156>

AB The factor that affect accurate, quant. results to be obtained by neutron-induced  $\gamma$ -ray emission tomog. are stated. The technique, which is a combination of NAA with computerized  $\gamma$ -ray emission tomog., would be enhanced by the use of multiple detector assemblies, in geometrical configurations, which simultaneously record the  $\gamma$ -rays emitted and improve detection efficiency. Developments in the past few years in positron emission tomog. where scanners made of single scintillation block detectors, cut into smaller segments, to form individual crystal detector elements and packed in ring around the radioactive object, are discussed. The coincident detection efficiency for annihilation photons and **cascade  $\gamma$ -rays** for such systems are considered and the possibilities of carrying out NIGET with coincident  $\gamma$ -ray tomog. were explored while indicating some of the limitations.

IT Scintillation detectors

(calculated detection efficiencies for coincident  $\gamma$ -rays for Bi germanate system with 16 block detectors)

IT Neutron activation analysis

(determination of the elemental distribution in sample using neutron-induced gamma-ray emission tomog.)

IT 10198-40-0, Cobalt-60, analysis 14265-76-0, Hafnium-179, analysis

14336-64-2, Cadmium-111, analysis

RL: ANT (**Analyte**); ANST (Analytical study)

(calculated detection efficiencies for coincident  $\gamma$ -rays for Bi germanate system with 16 block detectors)

RN 10198-40-0 HCPLUS

CN Cobalt, isotope of mass 60 (CA INDEX NAME)

$^{60}\text{Co}$

RN 14265-76-0 HCPLUS

CN Hafnium, isotope of mass 179 (CA INDEX NAME)

$^{179}\text{Hf}$

RN 14336-64-2 HCPLUS

CN Cadmium, isotope of mass 111 (CA INDEX NAME)

$^{111}\text{Cd}$

L128 ANSWER 8 OF 42 HCPLUS COPYRIGHT ACS on STN

AN 1996:369263 HCPLUS

DN 125:125256

OREF 125:23289a,23292a

ED Entered STN: 26 Jun 1996

TI Nuclide yields in ( $\gamma, \gamma'$ )-, (.gamma

.,n)- and ( $\gamma,p$ ) reactions at E<sub>max</sub> = 13.3 MeV

AU Kungurov, F. R.; Muminov, T. M.; Mukhamedov, A. K.; Saidmuradov, Zh.; Salikhbaev, U. S.; Safarov, A. N.

CS Samarkand State University, Samarkand, Uzbekistan

SO Izvestiya Akademii Nauk, Seriya Fizicheskaya (1996), 60(1), 201-205

CODEN: IRAFEO

PB Nauka

DT Journal

LA Russian

CC 70-1 (Nuclear Phenomena)

AB Nuclide yields in photonuclear reactions were determined at the electron accelerator of Samarkand University. The results are planned to be used in activation anal.

ST nuclide yield photonuclear reaction electron accelerator

IT Gamma ray

(yields of various nuclides in ( $\gamma,n$ )-, ( $\gamma$ ,p)- and ( $\gamma,\gamma'$ ) reactions on medium and heavy nuclei at gamma-ray maximum energy of 13.3 MeV)

IT 14336-64-2, Cadmium-111, processes

RL: PEP (Physical, engineering or chemical process); PROC (Process) (excitation of isomeric state of 111Cd in (

$\gamma,\gamma'$ ) inelastic scattering on)

RN 14336-64-2 HCPLUS

CN Cadmium, isotope of mass 111 (CA INDEX NAME)

111Cd

IT 14191-71-0, Indium-115, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(yields of metastable 114In and 115In in gamma-ray bombardment of)

RN 14191-71-0 HCPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

115In

L128 ANSWER 26 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1979:211284 HCAPLUS

DN 90:211284

OREF 90:33515a,33518a

TI **Excitation** of nuclear isomers by  $\gamma$  rays from cobalt-60

AU Watanabe, Yoshihisa; Mukoyama, Takeshi

CS Inst. Chem. Res., Kyoto Univ., Kyoto, Japan

SO Bulletin of the Institute for Chemical Research, Kyoto University (1979), 57(1), 72-82

CODEN: BICRAS; ISSN: 0023-6071

DT Journal

LA English

CC 70-2 (Nuclear Phenomena)

AB **Long-lived** isomeric states in stable nuclei  $^{111}\text{Cd}$  [ 14336-64-2] and  $^{115}\text{In}$  [14191-71-0] were **excited** by the Compton-scattered  $\gamma$ -rays from  $^{60}\text{Co}$ . The flux of the scattered photons at resonance was calculated in the single-scattering approximation. By using the observed  $\gamma$  activities of  $^{111m}\text{Cd}$  and  $^{115m}\text{In}$  and the calculated photon flux, the integral cross sections for the isomer production by photoexcitation of  $^{111}\text{Cd}$  and  $^{115}\text{In}$  were evaluated to be  $(3.5 \pm 0.4) \times 10^{-25}$  and  $(1.9 \pm 0.1) \times 10^{-25} \text{ cm}^2\text{-eV}$ , resp.

IT **Gamma ray**

(scattering of, by cadmium-111 and indium-115, isomer excitation in)

IT 14191-71-0, properties 14336-64-2, properties

RL: PRP (Properties)

(nuclear energy levels of, **excited** by  $\gamma$ -ray scattering, isomeric)

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

$^{115}\text{In}$

RN 14336-64-2 HCAPLUS

CN Cadmium, isotope of mass 111 (CA INDEX NAME)

$^{111}\text{Cd}$

L128 ANSWER 31 OF 42 HCPLUS COPYRIGHT ACS on STN

AN 1973:460156 HCPLUS

DN 79:60156

OREF 79:9657a,9660a

TI Nuclear quadrupole interaction studies by perturbed angular correlations

AU Haas, H.; Shirley, D. A.

CS Lawrence Berkeley Lab., Univ. California, Berkeley, CA, USA

SO Journal of Chemical Physics (1973), 58(8), 3339-55

CODEN: JCPSA6; ISSN: 0021-9606

DT Journal

LA English

CC 75-2 (Nuclear Phenomena)

Section cross-reference(s): 73

AB A comprehensive study was made of the applicability of .gamma .-ray angular correlations to the determination of quadrupole interactions in metals and insulating solids. Dynamic effects were studied in solns. and gases. A total of 14  $\gamma$ -ray cascades were employed. Several nuclear spins were confirmed and the quadrupole moments of 10 excited nuclear states were determined or estimated from the data. Quadrupole coupling consts. were determined for excited states of the following nuclei in metallic host lattices of the same element: 44Sc, 99Ru, 111Cd, 117In, 187Re, 199Hg. Coupling consts. were also measured for the following isotope (lattice) combinations: 99Ru(Zn, Cd, Sn, Sb), 100Rh(Zn, Ru, Cu5Zn3, Pd2Al, PdPb2), 111Cd(In, Hg, Tl, CdSb, Cd3Ag, Zn, Ga, In, Sn, Sb, Bi, AuIn, InBi, In2Bi), 115In(Cd), 117In(Cd, Sn), 131I(Te), 181Ta(HfB2, HfSi2), 204Pb(Cd, In, Sn, As, Sb, Bi, Hg, Tl, PdPb2). Systematic variations of e2qQ with host-lattice structure were observed and host and solute properties were found to be separable to some extent for nontransition metals. The nuclei 111Cd, 115In, 117In, 199Hg, and 204Pb were used to determine a total of 50 quadrupole coupling consts. in insulators, including 20 with non-zero asymmetry parameters, which give oscillatory but aperiodic correlation functions. It was strikingly (and exhaustively) demonstrated that good detns. of quadrupole coupling consts. could be made following isomeric transitions (with no elemental transmutation) and  $\beta$ -decay (with elemental transmutation). However, in no case was it possible to derive a coupling constant from a  $\gamma$ -ray cascade preceded directly by e-capture decay, presumably because the sudden creation of a K-hole, and the Auger and "shake-off" events that follow, destroy the chemical integrity of the species under study. Relaxation times were determined for a number of liquid samples. Studies of dimethyl-111Cd in various buffer gases showed that the spin memory was lost in 1 collision with heavy mols., but that light mols. required several collisions.

IT Gamma ray

(angular correlation of, quadrupole interactions and perturbed)

IT Nuclear energy level

(of atomic nuclei, quadrupole moments and spins of)

IT 14191-71-0, properties

RL: RCT (Reactant); RACT (Reactant or reagent)

(quadrupole coupling of, in cadmium and cadmium insulating compds.)

RN 14191-71-0 HCPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

115In

IT 14336-64-2, properties

RL: RCT (Reactant); RACT (Reactant or reagent)

(quadrupole interactions of)

RN 14336-64-2 HCPLUS

CN Cadmium, isotope of mass 111 (CA INDEX NAME)

111Cd

**L128 ANSWER** 33 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1970:138716 HCAPLUS

DN 72:138716

OREF 72:24773a,24776a

TI Determination of the amounts of elements in ore and products of their reprocessing by **excitation** of metastable **states** according to a ( $\gamma, \gamma'$ ) reaction

AU Kodiri, S.; Starchik, L. P.

CS USSR

SO Zavodskaya Laboratoriya (1970), 36(2), 191-5

CODEN: ZVDLAU; ISSN: 0321-4265

DT Journal

LA Russian

CC 76 (Nuclear Technology)

AB With the aid of the bremsstrahlung obtained from a 4.2-MeV electron linear accelerator (average output current of 50  $\mu$ A) metastable states of the following nuclides were **excited**: 77Se, 87Sr, 89Y, 103Rh, 107Ag, 109Ag, 111Cd, 113In, 115In, 117Sn, 135Ba, 137Ba, 176Lu, 179Hf, 195Pt, 197Au, and 199Hg. Either a 0.95-mm Pt target or a 2-mm Pb target was used. Time of the irradiation was of 3-4 **half-lives** for each element. The activity induced was measured by a  $\gamma$ -ray spectrometer with 40 + 50 mm NaI(Tl) crystal. Based on photopeak area detns. the following sensitivity limits (in mg) were found: 77Sem 0.16, 87Srm 0.3, 89Ym 3.4, 103Rhm 2.01, 107Agm 0.24, 111Cdm 0.08, 115Inm 0.024, 117Snm 80.0, 135Bam 7.1, 137Bam 1.9, 176L um 0.96, 179Hfm 0.1, 195Ptm 0.2, 197Aum 0.1, 199Hgm 0.9.

IT **Gamma** rays, chemical and physical effects

(radioactivation by)

IT Analysis

(radioactivation, by **gamma** rays)

IT 13981-59-4, analysis 14191-71-0, analysis

14191-88-9, analysis 14265-76-0, analysis

14336-64-2, analysis

RL: ANT (Analyte); ANST (Analytical study)

(determination of, by **gamma**-ray activation)

RN 13981-59-4 HCAPLUS

CN Tin, isotope of mass 117 (CA INDEX NAME)

117Sn

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

115In

RN 14191-88-9 HCAPLUS

CN Platinum, isotope of mass 195 (CA INDEX NAME)

195Pt

RN 14265-76-0 HCAPLUS

CN Hafnium, isotope of mass 179 (CA INDEX NAME)

179Hf

RN 14336-64-2 HCAPLUS

CN Cadmium, isotope of mass 111 (CA INDEX NAME)

111Cd

**L128 ANSWER** 40 OF 42 HCPLUS COPYRIGHT ACS on STN

AN 1951:46442 HCPLUS  
 DN 45:46442  
 OREF 45:7890a-b  
 TI Natural radioactivity of indium  
 AU Cohen, S. G.  
 CS Hebrew Univ., Jerusalem  
 SO Nature (London, United Kingdom) (1951), 167, 779  
 CODEN: NATUAS; ISSN: 0028-0836  
 DT Journal  
 LA Unavailable  
 CC 3A (Nuclear Phenomena)

**AB** **Natural radioactivity was sought in 2 pairs of neighboring isobars: In113-Cd113 and In115-Sn115.** By use of a counter with an In cathode, some activity was found. The radiation could not be identified as K **x -rays** of Cd; this fact indicates that any K-capture half **life** must be >10<sup>14</sup> yrs.  $\beta$ -Rays from the In have an energy of a few hundred e.kv., as determined from absorption measurements. In115 probably decays by  $\beta$ -emission to Sn115, with a half **life** of .apprx.10<sup>14</sup> yrs. By use of counters with different gas fillings, results were obtained which are consistent with the presence of a small amount of Cd L **x-rays**, which may arise after decay of In113 by L-capture, but evidence is inconclusive owing to the small counting rates involved. With L-capture, a half **life** of .apprx.10<sup>12</sup> yrs. is to be expected.

IT **14336-66-4**, Cadmium, isotope of mass 113  
 (from In115)  
 RN 14336-66-4 HCPLUS  
 CN Cadmium, isotope of mass 113 (CA INDEX NAME)

113Cd

IT **14191-71-0**, Indium, isotope of mass 115  
 (radioactivity of)  
 RN 14191-71-0 HCPLUS  
 CN Indium, isotope of mass 115 (CA INDEX NAME)

115In

**L128 ANSWER** 1 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 2005:636704 HCAPLUS  
 DN 144:75858  
 ED Entered STN: 22 Jul 2005  
 TI Investigation of formation of residual nuclei from natU by reactions with 660 MeV protons  
 AU Adam, J.; Katovsky, K.; Michel, R.; Balabekyan, A.; Tsoupko-Sitnikov, V. M.; Kalinnikov, V. G.; Pronskikh, V. S.; Solnyshkin, A. A.; Stegailov, V. I.; Mashnik, S. G.; Prael, R. E.; Gudima, K. K.  
 CS Joint Institute for Nuclear Research, Moscow, Russia  
 SO AIP Conference Proceedings (2005), 769(International Conference on Nuclear Data for Science and Technology, 2005, Part 1), 1043-1046  
 CODEN: APCPCS; ISSN: 0094-243X  
 PB American Institute of Physics  
 DT Journal  
 LA English  
 CC 70-1 (Nuclear Phenomena)  
 AB Thin natural uranium targets were irradiated by a 660 MeV proton beam from the Phasotron accelerator at the Joint Institute for Nuclear Research in Dubna, Russia. Cross-sections of the formation of residual nuclei natU(p,xpyn)AZ Res are determined by methods of gamma spectroscopy. Until now, 43 **long-lived** ( $T_{1/2} > 100$  days) isotopes were observed and their cross-sections determined. More than 350 intermediate-**lived** (1 day  $< T_{1/2} < 100$  days) and **short-lived** ( $T_{1/2} < 1$  day) isotopes have been identified in the  $\gamma$ -spectra and many unanalyzed lines yet remain. Final results for 43 **long-lived** isotopes and upper cross-section limits for 27 **long-lived** isotopes are presented in this paper and compared with results by five different models.  
 IT Nuclear reaction fragments  
     (formation of residual nuclei from natU by reactions with 660 MeV protons)  
 IT 10198-40-0P, Cobalt, isotope of mass 60, preparation  
 14265-77-1P, Hafnium, isotope of mass 178, preparation  
 14314-35-3P, Tin, isotope of mass 119, preparation  
 14336-66-4P, Cadmium, isotope of mass 113, preparation  
 378782-82-2P, **Niobium, isotope of mass 93m**(16.1 yr), preparation  
 RL: FMU (Formation, unclassified); PNU (Preparation, unclassified); FORM (Formation, nonpreparative); PREP (Preparation)  
     (formation of residual nuclei from natU by reactions with 660 MeV protons)  
 RN 10198-40-0 HCAPLUS  
 CN Cobalt, isotope of mass 60 (CA INDEX NAME)  

60Co

 RN 14265-77-1 HCAPLUS  
 CN Hafnium, isotope of mass 178 (CA INDEX NAME)  

178Hf

 RN 14314-35-3 HCAPLUS  
 CN Tin, isotope of mass 119 (CA INDEX NAME)  

119Sn

 RN 14336-66-4 HCAPLUS  
 CN Cadmium, isotope of mass 113 (CA INDEX NAME)  

113Cd

 RN 378782-82-2 HCAPLUS  
 CN **Niobium, isotope of mass 93m**(16.1 yr) (CA INDEX NAME)  

Nb



**L128 ANSWER** 2 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 2004:328024 HCAPLUS  
 DN 141:92038  
 ED Entered STN: 22 Apr 2004  
 TI Devgaon (H3) chondrite: classification and complex cosmic ray exposure history  
 AU Murty, S. V. S.; Rai, V. K.; Shukla, A. D.; Srinivasan, G.; Shukla, P. N.;  
 Suthar, K. M.; Bhandari, N.; Bischoff, A.  
 CS Planetary and Geosciences Division, Physical Research Laboratory,  
 Ahmedabad, 380009, India  
 SO Meteoritics & Planetary Science (2004), 39(3), 387-399  
 CODEN: MPSCFY; ISSN: 1086-9379  
 PB Meteoritical Society  
 DT Journal  
 LA English  
 CC 53-9 (Mineralogical and Geological Chemistry)  
 AB The Devgaon meteorite fell in India on Feb. 12, 2001 and was immediately collected. It is an ordinary chondrite having a number of SiO<sub>2</sub>-rich objects and some Ca, Al-rich inclusions. Olivines (Fa17-19) are fairly equilibrated, while pyroxenes (Fs4-20) are unequilibrated. Occasionally, shock veins are visible, but the bulk rock sample is very weakly shocked (S2). Chondrules and chondrule fragments are abundant. Based on chemical and petro. features, Devgaon is classified as an H3.8 group chondrite. Several cosmogenic radionuclides ranging in **half-lives** from 5.6 d (52Mn) to 7.3 + 105 yr (26Al), noble gases (He, Ne, Ar, Kr, and Xe), and particle track d. were measured. The track d. in olivines from 5 spot samples varies between (4.6 to 9) + 106 cm<sup>-2</sup> showing a small gradient within the meteorite. The light noble gases are dominated by cosmogenic and radiogenic components. Large amts. of trapped gases (Ar, Kr, and Xe) are present. In addition, (n,  $\gamma$ ) products from Br and I are found in Kr and Xe, resp. The average cosmic ray exposure age of 101 ± 8 Ma is derived based on cosmogenic 38Ar, 83Kr, and 126Xe. The track production rates correspond to shielding depths of .apprx.4.9 to 7.8 cm, indicating that the stone suffered type IV ablation. Low 60Co, high (22Ne/21Ne)c, and large neutron produced excesses at 80Kr, 82Kr, and 128Xe indicate a complex exposure history of the meteoroid. In the first stage, a meter-sized body was exposed for nearly 108 yr in the interplanetary space that broke up in .apprx.50 cm-sized fragments about a million years ago (stage 2), before it was captured by the Earth.  
 IT Radionuclides, occurrence  
 RL: GOC (Geological or astronomical occurrence); OCCU (Occurrence)  
 (cosmogenic; classification and complex cosmic ray exposure history of Devgaon (H3) chondrite)  
 IT 10198-40-0, Cobalt-60, occurrence 13965-99-6,  
 Xenon-129, occurrence 14683-11-5, Xenon-131, occurrence  
 RL: GOC (Geological or astronomical occurrence); OCCU (Occurrence)  
 (classification and complex cosmic ray exposure history of Devgaon (H3) chondrite)  
 RN 10198-40-0 HCAPLUS  
 CN Cobalt, isotope of mass 60 (CA INDEX NAME)  
 60Co  
 RN 13965-99-6 HCAPLUS  
 CN Xenon, isotope of mass 129 (CA INDEX NAME)  
 129Xe  
 RN 14683-11-5 HCAPLUS  
 CN Xenon, isotope of mass 131 (CA INDEX NAME)  
 131Xe

L128 ANSWER 3 OF 42 HCPLUS COPYRIGHT ACS on STN

AN 2003:146929 HCPLUS  
 DN 138:176694  
 ED Entered STN: 27 Feb 2003  
 TI Differential effective cross sections of photonuclear reactions induced by bremsstrahlung of 11.8 MeV electrons in a thick tungsten target  
 AU Abibullaev, N. A.; Begimkulov, Kh. Kh.; Salikhbaev, U. S.  
 CS Samark. Gos. Univ., Samarkand, Uzbekistan  
 SO O'zbekiston Fizika Jurnali (2002), 4(4), 277-282  
 CODEN: UJPZAR  
 PB Izdatel'stvo Fan  
 DT Journal  
 LA Russian  
 CC 70-1 (Nuclear Phenomena)  
 AB In this work, the method of spherical activation spectrometry was used to measure the energy spectra and angular distribution of the bremsstrahlung from the MT-22S microtron electrons. Using the exptl. determined spectra of bremsstrahlung radiation with  $E_{\gamma}(\text{max}) = 11.8 \text{ MeV}$  and the data on **excitation** functions, the effective cross sections for 63 photonuclear reactions were studied.  
 IT **Gamma** ray interactions  
     (differential effective cross section measurements of photonuclear reactions induced by bremsstrahlung of 11.8-MeV electrons in thick tungsten target)  
 IT Nuclear reactions  
     (photonuclear; differential effective cross section measurements of photonuclear reactions induced by bremsstrahlung of 11.8-MeV electrons in thick tungsten target)  
 IT **14191-71-0**, Indium-115, reactions  
 RL: PEP (Physical, engineering or chemical process); PYP (Physical process); RCT (Reactant); PROC (Process); RACT (Reactant or reagent)  
     (differential effective cross section measurements of **gamma**-ray inelastic scattering and neutron photoprodn. on  $^{115}\text{In}$  induced by bremsstrahlung of 11.8-MeV electrons in thick tungsten target)  
 RN 14191-71-0 HCPLUS  
 CN Indium, isotope of mass 115 (CA INDEX NAME)  
  
 $^{115}\text{In}$   
 IT **14336-64-2**, Cadmium-111, processes  
 RL: PEP (Physical, engineering or chemical process); PYP (Physical process); PROC (Process)  
     (differential effective cross section measurements of **gamma**-ray inelastic scattering on  $^{111}\text{Cd}$  with metastable **state** **excitation** induced by bremsstrahlung of 11.8-MeV electrons in thick tungsten target)  
 RN 14336-64-2 HCPLUS  
 CN Cadmium, isotope of mass 111 (CA INDEX NAME)  
  
 $^{111}\text{Cd}$   
 IT **14265-76-0**, Hafnium-179, processes  
 RL: PEP (Physical, engineering or chemical process); PYP (Physical process); PROC (Process)  
     (differential effective cross section measurements of **gamma**-ray inelastic scattering on  $^{179},^{180}\text{Hf}$  with metastable **state** **excitation** induced by bremsstrahlung of 11.8-MeV electrons in thick tungsten target)  
 RN 14265-76-0 HCPLUS  
 CN Hafnium, isotope of mass 179 (CA INDEX NAME)

$^{179}\text{Hf}$

**L128 ANSWER** 4 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1999:561975 HCAPLUS  
 DN 131:234316  
 ED Entered STN: 03 Sep 1999  
 TI Evaluation of a "best set" of average cross section measurements in the 235U(nth,f) neutron field  
 AU Mannhart, W.  
 CS Physikalisch-Technische Bundesanstalt Braunschweig, Braunschweig, Germany  
 SO Berichte des Forschungszentrums Juelich (1999), Juel-3660, 40-47  
 CODEN: FJBEE5; ISSN: 0366-0885  
 DT Report  
 LA English  
 CC 70-1 (Nuclear Phenomena)  
 AB Average cross sections measured in the thermal neutron-induced fission neutron field of 235U were evaluated with a total of 200 data. The evaluation comprised the measurement of 30 different neutron reactions. The results were given and compared with 2 previous evaluations. **X-ray** and  $\gamma$ -ray emission probabilities of 169Yb and the half-life of 153Sm were also studied.

IT **Gamma ray**  
**X-ray**  
 (x-ray and  $\gamma$ -ray emission probabilities of 169Yb)

IT 15117-96-1, Uranium-235, reactions  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (average cross section measurements in thermal neutron-induced fission neutron field of)

IT **10198-40-0**, Cobalt-60, formation (nonpreparative)  
 RL: FMU (Formation, unclassified); FORM (Formation, nonpreparative)  
 (average cross section measurements in thermal neutron-induced fission neutron field of 235U)

RN 10198-40-0 HCAPLUS  
 CN Cobalt, isotope of mass 60 (CA INDEX NAME)

$^{60}\text{Co}$

IT **14191-71-0**, Indium-115, reactions  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (average cross section measurements in thermal neutron-induced fission neutron field of 235U)

RN 14191-71-0 HCAPLUS  
 CN Indium, isotope of mass 115 (CA INDEX NAME)

$^{115}\text{In}$

**L128 ANSWER** 5 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1999:131580 HCAPLUS  
 DN 130:227082  
 ED Entered STN: 02 Mar 1999  
 TI Performance analysis for disposal of mixed low-level waste. II: Results  
 AU Waters, Robert D.; Gruebel, Marilyn M.  
 CS Sandia National Lab, Albuquerque, NM, 87185-1395, USA  
 SO Journal of Environmental Engineering (Reston, Virginia) (1999), 125(3), 258-266  
 CODEN: JOEEDU; ISSN: 0733-9372  
 PB American Society of Civil Engineers  
 AB Ten sites were evaluated as potential locations for disposal of mixed low-level radioactive waste using a simple methodol. Three environmental pathways (water, atmospheric, and inadvertent intruder) were analyzed using generic transport models that were modified for site-specific conditions. The results were summarized by grouping the 58 evaluated radionuclides according to their **half-lives** and environmental mobility and by their limiting pathway (i.e., the pathway providing the lowest permissible radionuclide concentration in disposed waste of the three evaluated pathways). The results indicate that all evaluated sites have the tech. capability for disposal of some radionuclides in the waste. For most radionuclides, the intruder scenarios were more important in determining permissible radionuclide concns. than the other pathways, particularly for arid sites. For humid sites, if the water pathway is not the most limiting and the permissible radionuclide concentration is high, a more sophisticated and rigorous anal. of the water pathway may not be warranted. However, if the permissible concentration is relatively low, more refined analyses may produce higher permissible concns. based on addnl. site characterization data.

IT Air pollution  
 Low-level radioactive wastes  
 Simulation and Modeling, physicochemical  
 (performance anal. for disposal of mixed low-level radioactive waste)  
 IT 10198-40-0, Cobalt-60, occurrence 14336-66-4,  
 Cadmium-113, occurrence 15726-30-4, Cesium-135, occurrence  
 RL: POL (Pollutant); OCCU (Occurrence)  
 (performance anal. for disposal of mixed low-level radioactive waste)  
 RN 10198-40-0 HCAPLUS  
 CN Cobalt, isotope of mass 60 (CA INDEX NAME)

<sup>60</sup>Co

RN 14336-66-4 HCAPLUS  
 CN Cadmium, isotope of mass 113 (CA INDEX NAME)

<sup>113</sup>Cd

RN 15726-30-4 HCAPLUS  
 CN Cesium, isotope of mass 135 (CA INDEX NAME)

<sup>135</sup>Cs

**L128 ANSWER** 9 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1995:601139 HCAPLUS  
 DN 123:67459  
 OREF 123:11869a,11872a  
 ED Entered STN: 09 Jun 1995  
 TI Photoexcitation of metastable states in  $^{111}\text{Cd}$ ,  $^{113}\text{In}$  and  $^{115}\text{In}$  in the 5-30 MeV energy range  
 AU Safar, J.; Lakosi, L.; Veres, A.; Sekine, T.; Yoshihara, K.  
 CS Institute Isotopes, Hungarian Academy Sciences, Budapest, H-1525, Hung.  
 SO Proc. Int. Symp. Capture Gamma-Ray Spectrosc. Relat. Top., 8th (1994), Meeting Date 1993, 629-31. Editor(s): Kern, Jean.  
 Publisher: World Sci., Singapore, Singapore.  
 CODEN: 60VEAP  
 DT Conference  
 LA English  
 CC 70-1 (Nuclear Phenomena)  
 AB The cross section of  $^{111}\text{Cd}(\gamma, \gamma')$   $^{111m}\text{Cd}$ ,  $^{113}\text{In}(\gamma, \gamma')$   $^{113m}\text{In}$  and  $^{115}\text{In}(\gamma, \gamma')$   $^{115m}\text{In}$  reactions has been studied in the 5-30 MeV **gamma**-energy region. Excitation functions have been calculated in the framework of a statistical  $\gamma$ -ray **cascade** model considering open nucleon emission channels and including preequil. contribution. At around the neutron separation energy, the calculated cross sections show reasonable agreement with the majority of exptl. data available in the literature. Above the  $(\gamma, n)$  threshold the integral cross section calculated for  $^{115}\text{In}$  agrees well with our recently measured value within the exptl. errors. An exptl. value of  $8.3 \pm 1.5$  mb MeV has been found for the  $^{113}\text{In}(\gamma, \gamma')$   $^{113m}\text{In}$  reaction at 15 MeV endpoint energy also in agreement with our calcns.  
 IT **Gamma ray**  
 (photoexcitation of metastable states in  $^{111}\text{Cd}$ ,  $^{113}\text{In}$  and  $^{115}\text{In}$ )  
 IT **14191-71-0**, Indium 115, properties **14336-64-2**,  
 Cadmium 111, properties  
 RL: PEP (Physical, engineering or chemical process); PRP (Properties);  
 PROC (Process)  
 (photoexcitation of metastable states in)  
 RN 14191-71-0 HCAPLUS  
 CN Indium, isotope of mass 115 (CA INDEX NAME)

$^{115}\text{In}$

RN 14336-64-2 HCAPLUS  
 CN Cadmium, isotope of mass 111 (CA INDEX NAME)

$^{111}\text{Cd}$

L128 ANSWER 10 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1993:416233 HCAPLUS  
 DN 119:16233  
 OREF 119:2941a,2944a  
 ED Entered STN: 10 Jul 1993  
 TI Nuclear isomer **excitation** cross section in ( $\gamma$ , .  
**gamma**.')<sub>m</sub> reactions at 4-15 MeV  
 AU Mazur, V. M.; Sokolyuk, I. V.; Bigan, Z. M.; Kobal, I. Yu.  
 CS Inst. Yad. Issled., Ukraine  
 SO Yadernaya Fizika (1993), 56(1), 20-5  
 CODEN: IDFZA7; ISSN: 0044-0027  
 DT Journal  
 LA Russian  
 CC 70-1 (Nuclear Phenomena)  
 AB Cross sections of  $\gamma$  inelastic scattering with **excitation** of isomer **states**  $\sigma_m$  of  $^{77}\text{Se}$ ,  
 $^{79}\text{Br}$ ,  $^{87}\text{Sr}$ , **111Cd**, **115In**, and  $^{137}\text{Ba}$  nuclei were measured at 4-15-MeV energies in  $\Delta E = 0.5$   
 MeV steps. For ( $\gamma, \gamma'$ )<sub>m</sub> reactions below the ( $\gamma, n$ ) thresholds the energy dependence of the  
 isomer ratios  $\eta = \sigma_m / \sigma_{tot}$  was considered. For nuclei whose spins of the ground and  
 isomer states differ by  $\Delta J = 4$ , the dependence  $\eta = f(A)$  is studied in a wide mass  
 interval  $77 < A < 197$ . Increase of  $\eta = f(A)$  with increase of  $A$  is in qual. agreement  
 with the statistical model predictions.  
 IT **Gamma ray**  
     (nuclear isomer **state excitation** by inelastic  
     scattering of)  
 IT **14191-71-0**, Indium-115, properties **14336-64-2**,  
 Cadmium-111, properties  
 RL: PRP (Properties)  
     (nuclear isomer **state excitation** in, by  
      $\gamma$ -ray inelastic scattering)  
 RN 14191-71-0 HCAPLUS  
 CN Indium, isotope of mass 115 (CA INDEX NAME)

$^{115}\text{In}$

RN 14336-64-2 HCAPLUS  
 CN Cadmium, isotope of mass 111 (CA INDEX NAME)

$^{111}\text{Cd}$

**L128 ANSWER** 11 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1990:162388 HCAPLUS  
 DN 112:162388  
 OREF 112:27387a,27390a  
 ED Entered STN: 28 Apr 1990  
 TI The Torino, H6, meteorite shower  
 AU Bhandari, N.; Bonino, G.; Callegari, E.; Castagnoli, G. Cini; Mathew, K.  
 J.; Padia, J. T.; Queirazza, G.  
 CS Ist. Cosmogeofis., Univ. Torino, Turin, Italy  
 SO Meteoritics (1989), 24(1), 29-34  
 CODEN: MERTAW; ISSN: 0026-1114  
 DT Journal  
 LA English  
 CC 53-9 (Mineralogical and Geological Chemistry)  
 AB A meteorite shower fell at Torino, Italy, on 18 May, 1988. Petrog. studies indicate that the stone is an H6 chondrite having features of moderate to severe shock. Chemical analyses of the meteorite are reported. Cosmic ray-produced  $^{3}\text{He}$ ,  $^{21}\text{Ne}$ , and  $^{126}\text{Xe}$  yield an exposure age of .apprx.48 Myr. The cosmic ray track densities in 3 fragments range between 1.8 + 105 to 5 + 105/cm<sup>2</sup> suggesting .apprx.99% mass ablation in the atmospheric. Twelve radioisotopes with **half lives** ranging between 5.6 days to 7.3 + 105 yr were measured with high precision (2-10%). Marginal signals were observed for several short-lived nuclides and upper limits were obtained for the activity levels of 8 radionuclides ( $^{24}\text{Na}$ ,  $^{48}\text{Cr}$ ,  $^{57}\text{Ni}$ ,  $^{47}\text{Sc}$ ,  $^{47}\text{Ca}$ ,  $^{59}\text{Fe}$ ,  $^{42}\text{Ar}$ , and  $^{44}\text{Ti}$ ), some of which have not been hitherto detected in fresh falls. The data are generally consistent with the nuclide production by galactic cosmic rays when modulation due to the solar cycle is taken into consideration. The preatm. radius of the chondrite was .apprx.20 cm, consistent with track densities and activity levels of  $^{60}\text{Co}$ ,  $^{26}\text{Al}$ , and other radionuclides.  
 IT Trace elements, occurrence  
 RL: OCCU (Occurrence)  
 (in chondrite meteorite, of Torino, Italy)  
 IT Fission fragments and products  
 RL: PRP (Properties)  
 (tracks of, d. of, in chondrite meteorite, ablation in atmospheric in relation to, of Torino, Italy)  
 IT 10198-40-0, Cobalt-60, occurrence 13965-99-6, Xenon-129, occurrence  
 14683-11-5, Xenon-131, occurrence  
 RL: OCCU (Occurrence)  
 (in chondrite meteorite, of Torino, Italy)  
 RN 10198-40-0 HCAPLUS  
 CN Cobalt, isotope of mass 60 (CA INDEX NAME)  
 60Co  
 RN 13965-99-6 HCAPLUS  
 CN Xenon, isotope of mass 129 (CA INDEX NAME)  
 129Xe  
 RN 14683-11-5 HCAPLUS  
 CN Xenon, isotope of mass 131 (CA INDEX NAME)  
 131Xe

**L128 ANSWER** 12 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1989:453313 HCAPLUS  
 DN 111:53313  
 OREF 111:8993a,8996a  
 ED Entered STN: 20 Aug 1989  
 TI Dose conversion factors for air, water, soil and building materials  
 AU Holford, R. M.  
 CS Chalk River Nucl. Lab., Chalk River, ON, K0J 1J0, Can.  
 SO At. Energy Can. Ltd., [Rep.] AECL (1988), AECL-9825, 449 pp.  
 CODEN: AECRAN; ISSN: 0067-0367  
 DT Report  
 LA English  
 CC 8-1 (Radiation Biochemistry)  
 Section cross-reference(s): 59, 71  
 AB This report provides dose conversion factors (DCFs) for 496 isotopes in 7 exposure situations. The tables include values for **gamma** dose at the body surface, beta dose at the body surface, beta dose to skin, DCFs for 24 individual organs, and an estimate of the DCF for the ED equivalent calculated according to ICRP rules. These calcns. were carried out in support of the Canadian Nuclear Waste-management Program, and are intended to form part of the assessment of the likely dose to humans from long- **lived** nuclides escaping from a disposal site.  
 IT **10198-40-0**, biological studies  
 RL: BIOL (Biological study)  
     (dosimetry of metastable and, in organs from building materials and environmental contamination)  
 RN 10198-40-0 HCAPLUS  
 CN Cobalt, isotope of mass 60 (CA INDEX NAME)  
  
<sup>60</sup>Co  
 IT **14191-71-0**, IN-115 element, biological studies  
 RL: ANT (Analyte); ANST (Analytical study)  
     (dosimetry of metastable and, in organs, from building materials and environmental contamination)  
 RN 14191-71-0 HCAPLUS  
 CN Indium, isotope of mass 115 (CA INDEX NAME)

<sup>115</sup>In

**L128 ANSWER** 13 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1989:413874 HCAPLUS  
 DN 111:13874  
 OREF 111:2385a,2388a  
 ED Entered STN: 08 Jul 1989  
 TI Nuclear activation techniques for measuring direct and backscattered components of intense bremsstrahlung pulses  
 AU Anderson, J. A.; Carroll, J. M.; Taylor, K. N.; Carroll, J. J.; Byrd, M. J.; Sinor, T. W.; Collins, C. B.; Agee, F. J.; Davis, D.; et al.  
 CS Cent. Quantum Electron., Univ. Texas, Dallas, Richardson, TX, USA  
 SO Nuclear Instruments & Methods in Physics Research, Section B: Beam Interactions with Materials and Atoms (1989), Volume Date 1988, B40-B41(2), 1189-92  
 CODEN: NIMBEU; ISSN: 0168-583X  
 DT Journal  
 LA English  
 CC 71-1 (Nuclear Technology)  
 AB High-voltage e accelerators used for bremsstrahlung generation can produce intense pulses of radiation with different endpoint energies. The energy spectrum can be changed by varying the charging voltage or by softening the photons with Compton scattering in a low atomic number material. The high dose rate and the flexible spectrum capabilities of the Aurora accelerator were used to investigate the potential for measuring the bremsstrahlung spectrum by photoactivation of nuclear isomeric states. Recent success in calibrating lower intensity sources has shown that gram-sized targets of isotopes, such as 115In, can be used to sample the incident **x- rays** at several discrete gateway energies. When irradiated at these energies the targets are **excited** to metastable **states** with lifetimes suitable for conventional counting after the flash.  
 IT Bremsstrahlung  
     (nuclear activation techniques for measuring direct and backscattered components of pulses of)  
 IT Accelerators  
     (electron, bremsstrahlung pulses from, nuclear activation techniques for measuring direct and backscattered components of)  
 IT **14336-64-2**, Cadmium-111, properties  
 RL: PRP (Properties)  
     (isomeric activation of metastable, in measurement of direct and backscattered components of bremsstrahlung pulses)  
 RN 14336-64-2 HCAPLUS  
 CN Cadmium, isotope of mass 111 (CA INDEX NAME)  
 111Cd  
 IT **14191-71-0**, Indium-115, properties  
 RL: PRP (Properties)  
     (photoactivation of, in measurement of direct and backscattered components of bremsstrahlung pulses)  
 RN 14191-71-0 HCAPLUS  
 CN Indium, isotope of mass 115 (CA INDEX NAME)  
 115In

**L128 ANSWER** 14 OF 42 HCPLUS COPYRIGHT ACS on STN

AN 1987:563018 HCPLUS  
 DN 107:163018  
 OREF 107:26081a,26084a  
 ED Entered STN: 31 Oct 1987  
 TI Spin-flip magnetic quadrupole transitions in tin-115 and tin-117  
 AU Fukuda, M.; Nagai, Y.; Irie, T.; Ejiri, H.  
 CS Dep. Phys., Osaka Univ., Toyonaka, 560, Japan  
 SO Nuclear Physics A (1987), A470(1), 1-12  
 CODEN: NUPABL; ISSN: 0375-9474  
 DT Journal  
 LA English  
 AB Electromagnetic properties of low-lying **excited states** in <sup>115</sup>Sn and <sup>117</sup>Sn were studied by means of in-beam and off-beam e. **gamma**. spectroscopy. Magnetic quadrupole (M2)  $\gamma$  matrix elements between the <sup>1</sup>h<sub>11/2</sub> and <sup>1</sup>g<sub>7/2</sub> quasi-n states in the Sn isotopes were reduced by factors g<sub>eff</sub>/g.**gamma** .free = 0.21-0.29 over single quasi-particle values. The reduction rates are about the same as those for analogous quasi-p transitions. There is, however, a finite difference between the renormalization factor g<sub>eff</sub>/g<sub>free</sub> for the M2  $\gamma$ -**decays** and those for the analogous 1st-forbidden  $\beta$ -**decays**.  
 IT Electron internal conversion  
     **Gamma** ray  
         (from tin isotopes, from deuteron bombardment of indium-115 and  
         **decay** of indium-117)  
 IT **14191-71-0**, Indium-115, reactions  
     RL: RCT (Reactant); RACT (Reactant or reagent)  
         (deuteron bombardment of, neutrons from)  
 RN 14191-71-0 HCPLUS  
 CN Indium, isotope of mass 115 (CA INDEX NAME)  
  
<sup>115</sup>In  
 IT **13981-59-4**, Tin-117, properties  
     RL: PRP (Properties)  
         (nuclear energy levels of, from **decay** of indium-117,  
         spin-flip magnetic quadrupole transitions of)  
 RN 13981-59-4 HCPLUS  
 CN Tin, isotope of mass 117 (CA INDEX NAME)

<sup>117</sup>Sn

**L128 ANSWER** 16 OF 42 HCPLUS COPYRIGHT ACS on STN

AN 1986:77103 HCPLUS  
 DN 104:77103  
 OREF 104:12143a,12146a  
 ED Entered STN: 08 Mar 1986  
 TI The strength of the spallation neutron flux of SINQ for radiation damage fusion technology  
 AU Hegedus, F.; Green, W. V.; Stiller, P.; Green, S.; Herrnberger, V.;  
 Stiefel, U.; Victoria, M.  
 CS Swiss Fed. Inst. Reactor Res., Wuerenlingen, 5303, Switz.  
 SO EIR-Ber. (1985), 579, 40 pp.  
 CODEN: EIBEDK; ISSN: 0531-6758  
 DT Report  
 LA English  
 CC 71-2 (Nuclear Technology)  
 Section cross-reference(s): 70  
 AB The SINQ spallation n source was simulated and its source strength measured in terms of the radiation damage parameters: total flux of spallation n with energy E <1.0 = MeV; spectral energy distribution; He and displacement damage production rates; and their ratio. This simulation used the p beam of the TRIUMF accelerator in Canada, and its molten Pb beam stop as the source of spallation n. The He production per p of beam was measured for several materials by vacuum extraction, inside a calibrated mass spectrograph; the spectrum, flux extraction inside a calibrated mass spectrograph; the spectrum, flux intensity, and the displacement radiation damage parameter were measured by multiple foil activation flux-unfolding combined with radiation damage calcns. The He/dpa (displacements per atom) ratio matches the fusion reactor 1st wall case; but the He production in Fe per mA-yr at a radial distance of 15 cm is estimated, to be 6 appm compared to 310 appm at the end of **life** in the 1st wall of NET.

IT Radiation, chemical and physical effects  
 (damage by, of fusion reactor first wall materials)

IT Nuclear fusion reactors  
 (first walls, radiation damage to, spallation neutron flux in relation to)

IT **10198-40-0P**, preparation

RL: PREP (Preparation)  
 (production of, by neutron bombardment of copper or nickel, **gamma** ray activities of product nuclei in relation to)

RN 10198-40-0 HCPLUS

CN Cobalt, isotope of mass 60 (CA INDEX NAME)

<sup>60</sup>Co

IT **14191-71-0P**, preparation

RL: PREP (Preparation)  
 (production of, by neutron bombardment of indium, **gamma** ray activities of product nuclei from)

RN 14191-71-0 HCPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

<sup>115</sup>In

**L128 ANSWER** 17 OF 42 HCPLUS COPYRIGHT ACS on STN

AN 1985:156656 HCPLUS  
 DN 102:156656  
 OREF 102:24529a,24532a  
 ED Entered STN: 04 May 1985  
 TI Long-lived radionuclides in low-level waste  
 AU Cline, James E.; Coe, Larry J.  
 CS Sci. Appl., Inc., USA  
 SO Proc. Annu. Participants' Inf. Meet.: Low-Level Waste Manage. Program  
     (1983), Issue 8308106, DE84 004134, 404-116 Publisher: NTIS,  
     Springfield, Va.  
 CODEN: 53ITAD  
 DT Conference  
 LA English  
 CC 71-11 (Nuclear Technology)  
 AB The radionuclide contents were studied of LWR-generated low-level waste. The objectives of the study are: (1) to analyze, using verified techniques, 150 archived samples for specified  $\beta$ - and  $\gamma$ -ray-emitting nuclides that had not previously been analyzed; (2) to analyze 20 new samples obtained from operating plants for all relevant nuclides and compare them to previous data to ascertain trends; (3) to develop empirical scaling factors through the use of which concns. of hard-to-analyze nuclides can be estimated from analyses of the  $\gamma$ -ray emitting nuclides. The new samples are analyzed and the results are summarized and interpreted. Scaling factor development is discussed. Factors are presented that relate 63Ni and 59Ni to 60Co for PWRs and to 58Co for BWRs, 90Sr to 137Cs for BWRs and 241Pu, 239Pu, 241Am, and 244Cm to 144Ce for all LWRs.  
 ST radioelement radioactive waste LWR; scaling factor radioelement  
     radioactive waste  
 IT Radioactive wastes  
     (low-level, determination of long-lived radionuclides in, from LWR,  
     scaling factors for)  
 IT 10198-40-0, analysis 15726-30-4, analysis  
     RL: ANT (Analyte); ANST (Analytical study)  
     (determination of, in low-level radioactive wastes)  
 RN 10198-40-0 HCPLUS  
 CN Cobalt, isotope of mass 60 (CA INDEX NAME)  
  
 $^{60}\text{Co}$   
 RN 15726-30-4 HCPLUS  
 CN Cesium, isotope of mass 135 (CA INDEX NAME)  
  
 $^{135}\text{Cs}$

**L128 ANSWER** 18 OF 42 HCPLUS COPYRIGHT ACS on STN

AN 1982:188345 HCPLUS  
 DN 96:188345  
 OREF 96:30941a,30944a  
 TI Nuclear data for **x-** or gamma-**ray** spectrometer efficiency calibrations  
 AU Hoppes, D. D.; Hutchinson, J. M. R.; Schima, F. J.; Unterweger, M. P.  
 CS Natl. Bur. Stand., Washington, DC, USA  
 SO NBS Special Publication (United States) (1982), 626, 85-99  
 CODEN: XNBSAV; ISSN: 0083-1883  
 DT Journal  
 LA English  
 CC 70-1 (Nuclear Phenomena)  
 Section cross-reference(s): 71, 73  
 AB **Half-lives** measured with reference ionization chambers are reported for 22Na, 24Na, 46Sc, 51Cr, 54Mn, 57Co, 58Co, 59Fe, 60Co, 65Zn, 67Ga, 75Se, 85Sr, 88Y, 99Mo, 99mTc (two chemical forms), 109Cd, 110mAg, 111In, 113Sn, 123I, 125I, 125Sb, 127Xe, 131I, 131mXe, 133Xe, 133Ba, 134Cs, 137Cs, 139Ce, 140Ba, 140La, 141Ce, 144Ce, 152Eu, 153Gd, 154Eu, 155Eu, 169Yb, 181W, 192Ir, 195Au, 198Au, 201Tl, 203Hg, 203Pb, 207Bi, and 228Th.  $\gamma$ -Ray probabilities per decay measured with calibrated Ge spectrometry systems are given for 88Y, 99Mo, 125Sb, 131I, 154Eu, and 228Th. The  $\gamma$ -**ray** to K-**x-ray** emission-rate ratio for 109Cd and the  $\gamma$ -ray probability per decay for 241Am as measured with NaI(Tl) systems of known efficiency are given.  
 IT **X-ray** spectrometry  
     (data for calibrations in)  
 IT Gamma ray  
     (spectrometry, data for calibrations in)  
 IT Nuclear spectrometers  
     (gamma-ray, efficiency calibration of germanium)  
 IT **14683-11-5**, reactions  
     RL: RCT (Reactant); RACT (Reactant or reagent)  
         (decay of metastable, half-**life** and  $\gamma$ -rays from)  
 RN 14683-11-5 HCPLUS  
 CN Xenon, isotope of mass 131 (CA INDEX NAME)

131Xe

IT **10198-40-0**, reactions  
     RL: RCT (Reactant); RACT (Reactant or reagent)  
         (decay of, half-**life** and  $\gamma$ -rays from)  
 RN 10198-40-0 HCPLUS  
 CN Cobalt, isotope of mass 60 (CA INDEX NAME)

60Co

**L128 ANSWER** 19 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1982:76244 HCAPLUS  
 DN 96:76244  
 OREF 96:12427a,12430a  
 TI Decay calculations on medium-level and actinide-containing wastes from the LWR fuel cycle. Part I. Basic data evaluation including activity and thermal decay power  
 AU Haug, H. O.  
 CS Inst. Heisse Chem., Kernforschungszent. Karlsruhe, Karlsruhe, Fed. Rep. Ger.  
 SO Kernforschungszent. Karlsruhe, [Ber.] KFK (1981), KFK 3221, 71 pp.  
 CODEN: KKBRAY; ISSN: 0303-4003  
 DT Report  
 LA German  
 CC 71-11 (Nuclear Technology)  
 Section cross-reference(s): 55, 60  
 AB A number of basic data on medium-level and actinide-containing waste streams from the LWR fuel cycle were evaluated, and the activity and thermal decay power were calculated for the nuclide inventories of cladding hulls and fuel assembly structural materials, for feed clarification sludge, medium-level aqueous process waste, low-level solid transuranium waste and for medium-level reactor operating waste. The activity as a function of decay time of the medium-level wastes decreases within 500 to 600 yr by 1-3 orders of magnitude and is at the same time .apprx.1-2 orders of magnitude lower than the activity of the high-level waste. The thermal decay power of the medium-level wastes decreases after 10-100 yr by .apprx.3 orders of magnitude and is about a factor of 10-100 less than that of high-level waste. In the very long term, the residual activity (and thermal power) decreases only slowly due to the long **half-lives** of the dominant actinides. The activity after >1000 yr is .apprx. 1-2 orders of magnitude lower than that of high-level waste, the low-level transuranium waste by a factor 10 to 4, resp. The activity per unit volume of the packaged waste of the medium-level and actinide-containing wastes, owing to the bigger volume of the conditioned wastes, is lower by 2-4 orders of magnitude up to .apprx.500 yr. After >1000 yr the activities per unit volume are lower by a factor of 20-200 than that of high-level waste.  
 IT Radioactive wastes  
     (actinide-containing and medium-level, from LWR fuel cycle, decay calcns. on)  
 IT Actinides  
     Fission fragments and products  
     RL: PRP (Properties)  
         (activity of, in aqueous medium-level radioactive wastes)  
 IT Nuclear reactor fuels and fuel elements  
     (cycles, plutonium estimated losses in LWR)  
 IT **14314-35-3**, properties **14390-73-9**, properties  
     RL: PRP (Properties)  
         (nuclear decay calcns. on metastable, in radioactive waste)  
 RN 14314-35-3 HCAPLUS  
 CN Tin, isotope of mass 119 (CA INDEX NAME)  

119Sn

 RN 14390-73-9 HCAPLUS  
 CN Tellurium, isotope of mass 125 (CA INDEX NAME)  

125Te

 IT **10198-40-0**, properties  
     RL: PRP (Properties)  
         (nuclear decay calcns. on radioactive wastes containing)  
 RN 10198-40-0 HCAPLUS  
 CN Cobalt, isotope of mass 60 (CA INDEX NAME)  

60Co

L128 ANSWER 20 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1981:522261 HCAPLUS

DN 95:122261

OREF 95:20397a,20400a

TI Beta decay rates for s-process studies

AU Cosner, Kenneth; Truran, James W.

CS Dep. Astron., Univ. Illinois, Urbana, IL, USA

SO Astrophysics and Space Science (1981), 78(1), 85-94

CODEN: APSSBE; ISSN: 0004-640X

DT Journal

LA English

CC 70-5 (Nuclear Phenomena)

AB The rates for a variety of  $\beta$  decay processes were determined as a function of temperature for nuclei which can participate in the s-process production of heavy elements, occurring in the presence of the  $^{22}\text{Ne}(\alpha, n)^{25}\text{Mg}$  source operating in the convective He shells of thermally pulsing stars. Specifically, calculated **half-lives** are presented for e- emission, e+ emission, and e- capture over the temperature range 108-109 K.

IT Stars

(nucleosynthesis in, slow neutron capture, rate calcns. for  $\beta$  decay processes as function of temperature in relation to)

IT Atomic nuclei

(nucleosynthesis of, rate calcns. for  $\beta$  decay processes as function of temperature in relation to s-process)

IT Beta decay

Electron-capture decay

(rate calcns. for, as function of temperature, s-process nucleosynthesis in relation to)

IT 10198-40-0, reactions 14191-71-0, reactions

RL: PRP (Properties)

(beta decay **half-life** of, temperature dependence of, s-process nucleosynthesis in relation to)

RN 10198-40-0 HCAPLUS

CN Cobalt, isotope of mass 60 (CA INDEX NAME)

$^{60}\text{Co}$

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

$^{115}\text{In}$

**L128 ANSWER** 21 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1981:146578 HCAPLUS  
 DN 94:146578  
 OREF 94:23893a,23896a

TI High resolution  $\gamma$  spectra of 40-44 MeV photon activation products. Part 3. A summary of  $\gamma$  rays, radionuclides and nuclear interferences observed

AU Williams, D. R.; Hislop, J. S.  
 CS Environ. Med. Sci. Div., AERE, Harwell, UK  
 SO U. K. At. Energy Res. Estab., Rep. (1980), AERE-R 9022, 32 pp.  
 CODEN: UKRGAL; ISSN: 0436-9734

DT Report  
 LA English  
 CC 70-2 (Nuclear Phenomena)  
 Section cross-reference(s): 79

AB A table of  $\gamma$ -rays observed in the high resolution **gamma** ray spectra of 40-44 MeV  $\gamma$  photon activation products is presented. This table is arranged in order of increasing **gamma** -ray energy and the parent isotopes, their **half-lives** and their inactive precursors are identified. Nuclear interferences caused by production of an active isotope from different parent elements have been identified and evaluated quant. These are also tabulated.

ST activation analysis **gamma** table  
 IT **Gamma** ray  
     (tables of)  
 IT Radiochemical analysis  
     (activation, **gamma**-ray tables for)  
 IT 10198-40-0, uses and miscellaneous 14191-71-0, uses  
     and miscellaneous  
 RL: USES (Uses)  
     (nuclear interference of, in activation anal.)  
 RN 10198-40-0 HCAPLUS  
 CN Cobalt, isotope of mass 60 (CA INDEX NAME)

60Co

RN 14191-71-0 HCAPLUS  
 CN Indium, isotope of mass 115 (CA INDEX NAME)

115In

**L128 ANSWER** 22 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1980:223010 HCAPLUS  
 DN 92:223010  
 OREF 92:35971a,35974a  
 TI Dose-rate conversion factors for external exposure to photon and electron radiation from radionuclides occurring in routine releases from nuclear fuel cycle facilities  
 AU Kocher, D. C.  
 CS Health Saf. Res. Div., Oak Ridge Natl. Lab., Oak Ridge, TN, 37830, USA  
 SO Health Physics (1980), 38(4), 543-621  
 CODEN: HLTPAO; ISSN: 0017-9078  
 DT Journal  
 LA English  
 CC 71-9 (Nuclear Technology)  
 Section cross-reference(s): 8  
 AB Dose-rate conversion factors for external exposure to photon and e radiation were calculated for 240 radionuclides of potential importance in routine releases from nuclear fuel cycle facilities. Exposure modes considered are immersion in contaminated air, immersion in contaminated water, and irradiation from a contaminated ground surface. For each exposure mode, dose-rate conversion factors for photons and e were calculated for tissue-equivalent material at the body surface of an exposed individual. Dose-rate conversion factors for photons only were calculated for 22 body organs.  
 IT Actinides  
 Fission fragments and products  
 Radioelements, biological studies  
 RL: BIOL (Biological study)  
 (dose rate at body surface and dose-conversion factors for various organs exposed to)  
 IT Nuclear reactor fuel reprocessing  
 (dose-rate conversion factors for external exposure to radiation from radioelements released in)  
 IT Gamma ray  
 (dose-rate conversion factors for, for various radioelements)  
 IT Dosimetry  
 (dose-rate conversion factors, for radioelements occurring in routine releases from fuel reprocessing facilities)  
 IT 10198-40-0, biological studies 14191-71-0, biological studies  
 RL: BIOL (Biological study)  
 (dose rate at body surface and dose-conversion factors for various organs exposed to)  
 RN 10198-40-0 HCAPLUS  
 CN Cobalt, isotope of mass 60 (CA INDEX NAME)  
 60Co  
 RN 14191-71-0 HCAPLUS  
 CN Indium, isotope of mass 115 (CA INDEX NAME)

115In

**L128 ANSWER** 23 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1980:157197 HCAPLUS

DN 92:157197

OREF 92:25355a,25358a

TI A consistent set of nuclear-parameter values for absolute INAA

AU Heft, Robert E.

SO DOE Symposium Series (1979), Volume Date 1978, 49(Comput. Act.

Anal. Gamma-Ray Spectrosc.), 495-510

CODEN: DOESD6; ISSN: 0164-2022

AB **Gamma** spectral anal. of irradiated material can be used to determine absolute disintegration rates for specific radionuclides. These data, together with measured values for the thermal and epithermal neutron fluxes and irradiation, cooling, and counting-time values, are all the exptl. information required to do absolute instrumental neutron activation anal. (INAA). The calcns. required to go from product photon emission rate to target nuclide amount depend on values used for the thermal-neutron-capture cross section, the resonance absorption integral, the **half-life**, and photon branching ratios. Values for these parameters were determined by irradiating and analyzing a series of elemental stds. The results of these measurements were combined with values reported by other workers to arrive at a set of recommended values for the consts. Values for 114 nuclides are listed.

IT Radiochemical analysis

(neutron activation, instrumental, nuclear parameter values for absolute)

IT 13981-59-4, properties 14191-71-0, properties 14191-88-9, properties 14336-64-2,

RL: PRP (Properties)

(**gamma** rays from metastable, following neutron capture in activation anal., consts. for)

RN 13981-59-4 HCAPLUS

CN Tin, isotope of mass 117 (CA INDEX NAME)

117Sn

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

115In

RN 14191-88-9 HCAPLUS

CN Platinum, isotope of mass 195 (CA INDEX NAME)

195Pt

RN 14336-64-2 HCAPLUS

CN Cadmium, isotope of mass 111 (CA INDEX NAME)

111Cd

IT 13967-67-4, properties 14191-71-0, properties 14191-88-9, properties 14265-76-0,  
(neutron cross sections of, for activation anal.)

RN 13967-67-4 HCAPLUS

CN Iridium, isotope of mass 193 (CA INDEX NAME)

193Ir

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

115In

RN 14191-88-9 HCAPLUS

CN Platinum, isotope of mass 195 (CA INDEX NAME)

195Pt

RN 14265-76-0 HCAPLUS

CN Hafnium, isotope of mass 179 (CA INDEX NAME)

179Hf

L128 ANSWER 24 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1980:48454 HCAPLUS

DN 92:48454

OREF 92:7927a,7930a

TI **Excitation** of indium-115, indium-113, and cadmium-111 nuclei in the photo-free annihilation of positrons

AU Vishnevskii, I. N.; Zheltonozhskii, V. A.; Svyato, V. P.; Trishin, V. V.

CS Inst. Yad. Issled., Leningrad, USSR

SO Izvestiya Akademii Nauk SSSR, Seriya Fizicheskaya (1979), 43(10), 2142-7

CODEN: IANFAY; ISSN: 0367-6765

DT Journal

LA Russian

CC 70-2 (Nuclear Phenomena)

AB **Excitation** of the isomeric levels of 336.2 keV in  $^{115}\text{In}$  [14191-71-0], 391.7 keV in  $^{113}\text{In}$  [14885-78-0], and 150.8 and 245.4 keV in  $^{111}\text{Cd}$  [14336-64-2] in photon-free  $e^+$  annihilation was studied. Natural In or Cd targets together with a planar  $^{64}\text{Cu}$   $e^+$  sources were in a sandwich-type arrangement. The **deexcitation . gamma**.-rays of the above levels was measured with Ge(Li) detectors. The cross sections of the level **excitation** was determined Direct **excitation** of the above levels with  $e^+$  and **excitation** by  $\gamma$ -rays was excluded. The levels at 1077.8 keV in  $^{115}\text{In}$  and 1129.4 keV in  $^{113}\text{In}$  were **excited** by the photon-free positron annihilation and the levels at 336.2 and 391.7 keV, resp., were populated by the **deexcitation** of the higher levels. The level populating the states at 150.8 and 245.4 keV in  $^{111}\text{Cd}$  is not known. The cross sections agree with those of T. Mikoyama et al. (1972) but do not agree with theor. predictions of D. P. Grechukhin et al. (1977).

IT **Gamma ray**

(from cadmium-111 and indium isotope level **deexcitation**)

IT 14191-71-0, properties 14336-64-2, properties

RL: PRP (Properties)

(nuclear energy level **excitation** of, by positron annihilation)

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

$^{115}\text{In}$

RN 14336-64-2 HCAPLUS

CN Cadmium, isotope of mass 111 (CA INDEX NAME)

$^{111}\text{Cd}$

**L128 ANSWER** 25 OF 42 HCPLUS COPYRIGHT ACS on STN

AN 1979:547062 HCPLUS

DN 91:147062

OREF 91:23615a,23618a

TI The 852 keV first-forbidden beta transition in cadmium-115g

AU Lakshminarayana, S.; Rao, M. Srinivasa; Rao, V. Seshagiri; Sastry, D. L.

CS Lab. Nucl. Res., Andhra Univ., Waltair, India

SO Nuovo Cimento della Societa Italiana di Fisica, A: Nuclei, Particles and Fields (1979), 52A(1), 92-8

CODEN: NIFAAM; ISSN: 0369-4097

DT Journal

LA English

CC 70-2 (Nuclear Phenomena)

AB  $\beta\gamma$  Angular-correlation expts. were performed on the 852 keV  $\beta$ -261 keV  $\gamma$  cascade in the ground-state decay of  $^{115}\text{Cd}$  [14336-68-6] by using a conventional fast-slow coincidence set-up. The present integral-correlation experiment confirmed a 3/2- spin-parity assignment for the 597-keV level in  $^{115}\text{In}$  [14191-71-0], while the differential-correlation measurements revealed a small energy dependence of the correlation function. The nonunique 1st-forbidden 852-keV  $\beta$  transition does not obey the  $\xi$ -approximation

IT Gamma ray  
(of indium-115, from decay of cadmium-115)

IT 14336-68-6, reactions

RL: PRP (Properties); RCT (Reactant); RACT (Reactant or reagent)  
(decay of, gamma-ray spectrum in)

IT 12587-47-2P

RL: PREP (Preparation)  
(from cadmium-115 decay)

IT 14191-71-0, properties

RL: PRP (Properties)  
(nuclear energy levels of, from cadmium-115 decay)

**L128 ANSWER** 27 OF 42 HCPLUS COPYRIGHT ACS on STN

AN 1979:111154 HCPLUS  
 DN 90:111154  
 OREF 90:17445a,17448a  
 TI **Nuclear excitation of indium-115 by positron annihilation with K-shell electrons**  
 AU Watanabe, Yoshihisa; Mukoyama, Takeshi; Shimizu, Sakae  
 CS Inst. Chem. Res., Kyoto Univ., Kyoto, Japan  
 SO Physical Review C: Nuclear Physics (1979), 19(1), 32-7  
 CODEN: PRVCAN; ISSN: 0556-2813  
 DT Journal  
 LA English  
 CC 70-2 (Nuclear Phenomena)  
 AB When a e+ annihilates with a K-shell e-, the excess energy liberated may be given to the nucleus involved. When this energy is just the right energy for any **excited state** within its level width of a fraction of an eV, nuclear excitation can be expected, and if the excited level **cascades** down to an isomeric state with appreciable lifetime, the occurrence of this annihilation mode can be confirmed by observing  $\gamma$ -transitions or conversion e- from this isomeric level. **Gamma-rays from  $^{115m}\text{In}$  [14191-71-0]** were observed after irradiation of natural In foils by e+ from  $^{64}\text{Cu}$ . By using the observed induced  $\gamma$  activity of  $^{115m}\text{In}$  and assuming this phenomenon to be a 2-step process, the cross sections of nuclear excitation by e+ annihilation for the 1078- and 1464-keV levels of  $^{115}\text{In}$  were evaluated as  $(3.9 \pm 1.4) + 10^{-24} \text{ cm}^2$  and  $(1.4 \pm 0.5) + 10^{-22} \text{ cm}^2$ , resp.  
 IT 12585-85-2  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (annihilation of, with K-shell electrons, cross sections for excitation  
 of indium-115 levels by)  
 IT 14191-71-0, properties  
 RL: PRP (Properties)  
 (nuclear energy levels of, excited by positron annihilation with  
 K-shell electrons, cross sections for)

**L128 ANSWER** 28 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1976:96564 HCAPLUS

DN 84:96564

OREF 84:15679a,15682a

TI Determination of the sign and magnitude of the nuclear quadrupole interaction by  $\beta$ - $\gamma$  directional correlations

AU Raghavan, R. S.; Raghavan, P.; Kaufmann, E. N.

CS Bell Lab., Murray Hill, NJ, USA

SO Physical Review C: Nuclear Physics (1975), 12(6), 2022-32  
CODEN: PRVCAN; ISSN: 0556-2813

DT Journal

LA English

CC 70-2 (Nuclear Phenomena)

AB Expts. are reported which demonstrate that the sign and magnitude of the quadrupole interaction of **excited** nuclear **states** can be reliably determined by means of time-differential  $\beta$ -.**gamma** . directional correlations with radioactive sources embedded in noncubic single crystals. The 828-keV level of  $^{115}\text{In}$  [**14191-71-0**] and the 247-keV level of  $^{111}\text{Cd}$  [**14336-64-2**], fed by an allowed and a unique first-forbidden  $\beta$ -**decay**, resp., were investigated by this method. The coupling consts.  $e2qQ/h$  of these levels in Cd metal at  $298^\circ\text{K}$ , are -146(5) MHz for  $^{115}\text{In}$  and +125(4) MHz for  $^{111}\text{Cd}$ . The pos. coupling constant for  $^{111}\text{Cd}$  in Cd coupled with a pos. quadrupole moment Q of this level implies a pos. elec. field gradient in Cd confirming recent theor. predictions. The neg.  $e2qQ$  for the  $^{115}\text{In}$  level in Cd metal correspondingly indicates a neg. Q for this level in conformity with its description as a member of a  $K = 1/2$  rotational band. A discussion is included of the major concepts underlying the various methods available now for determination of the sign of  $e2qQ$ . Also a detailed account of theor. formulas necessary to evaluate allowed and forbidden  $\beta$ -.**gamma** . perturbed angular correlations and a brief discussion of the significance of the technique for further studies of nuclear quadrupole effects in materials are given.

IT **Gamma ray**

(angular correlations of, with  $\beta$  particles, nuclear quadrupole interactions from)

IT Nuclear energy level

(of cadmium-111 and indium-115, quadrupole interactions of)

IT **14191-71-0, properties**

**RL: PRP (Properties)**  
(nuclear energy levels of, from decay of metastable cadmium-115, quadrupole interactions of)

RN **14191-71-0 HCAPLUS**

CN Indium, isotope of mass 115 (CA INDEX NAME)

$^{115}\text{In}$

IT **14336-64-2, properties**

**RL: PRP (Properties)**  
(nuclear energy levels of, from silver-111 decay, quadrupole interactions of)

RN **14336-64-2 HCAPLUS**

CN Cadmium, isotope of mass 111 (CA INDEX NAME)

$^{111}\text{Cd}$

**L128 ANSWER** 29 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1975:585169 HCAPLUS  
 DN 83:185169  
 OREF 83:29019a,29022a  
 TI **Short-lived** isotopes for use in nondestructive activation analysis  
 AU Bode, P.; De Bruin, M.; Korthoven, P. J. M.  
 CS Interuniv. React. Inst., Delft, Neth.  
 SO Journal of Radioanalytical Chemistry (1975), 26(1), 209-13  
 CODEN: JRACBN; ISSN: 0022-4081  
 DT Journal  
 LA English  
 CC 71-8 (Nuclear Technology)  
 Section cross-reference(s): 79  
 AB To obtain reliable data on **short-lived** isotopes for use in thermal neutron activation anal., expts. were carried out by using a fast rabbit transfer system. **Half-lives** of 28 **short-lived** isotopes were measured by using a counting system with a fixed dead-time. A Ge(Li) spectrometry system was used to determine the most important  $\gamma$ -ray energies and intensities of these isotopes. For the **half-lives**, an accuracy of better than 1% was attained, whereas for the  $\gamma$ -ray energies an accuracy of 0.1 keV was attained.  
 IT Radioelements, reactions  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (decay of, **half-lives** for)  
 IT Gamma ray  
 (from radioelements of interest in activation anal.)  
 IT Radiochemical analysis  
 (neutron activation, **short-lived** radioelement  
**halflife** for)  
 IT 14265-76-0, reactions 14265-77-1, reactions  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (decay of metastable, **half-life** for)  
 RN 14265-76-0 HCAPLUS  
 CN Hafnium, isotope of mass 179 (CA INDEX NAME)

$^{179}\text{Hf}$

RN 14265-77-1 HCAPLUS  
 CN Hafnium, isotope of mass 178 (CA INDEX NAME)

$^{178}\text{Hf}$

IT 14191-71-0, reactions 14265-77-1, reactions  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (neutron capture by,  $\gamma$ -rays from)  
 RN 14191-71-0 HCAPLUS  
 CN Indium, isotope of mass 115 (CA INDEX NAME)

$^{115}\text{In}$

RN 14265-77-1 HCAPLUS  
 CN Hafnium, isotope of mass 178 (CA INDEX NAME)

$^{178}\text{Hf}$

L128 ANSWER 30 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1974:54701 HCAPLUS

DN 80:54701

OREF 80:8863a,8866a

TI Gamma ray-multiplicity measurements for ( $\alpha$ ,p), ( $\alpha$ , $\alpha'$ ),  
(d, $\alpha$ ), (p, $\alpha$ ), (d,p), and (p,p') reactions

AU Degnan, J. H.; Cohen, B. L.; Rao, G. R.; Chan, K. C.; Shabason, L.

CS Nucl. Phys. Lab., Univ. Pittsburgh, Pittsburgh, PA, USA

SO Physical Review C: Nuclear Physics (1973), 8(6), 2255-66

CODEN: PRVCAN; ISSN: 0556-2813

DT Journal

LA English

CC 75-2 (Nuclear Phenomena)

AB By using a method introduced previously, measurements were made of the  $\gamma$ -ray multiplicity  $N_{\gamma}$  (the average number of  $\gamma$ -rays emitted in the **decay** of residual nuclei left by nuclear reactions) as a function of **excitation** energy  $E^*$  for (p, $\alpha$ ) and (d, $\alpha$ ) reactions on 51V, 56Fe, and 57Fe targets; ( $\alpha$ ,p) reactions on 51V, 56Fe, 57Fe, 58Ni, 64Ni, 93Nb, and Ag; ( $\alpha$ , $\alpha'$ ) reactions on 51V, 55Mn, 56Fe, 57Fe, 59Co, and 64Ni; (d,p) reactions on 51V, 55Mn, 56Fe, Ag, and 119Sn; and (p,p') reactions on 56Fe, 112Sn, and 122Sn. Bombarding energies ranged from 12 to 19 MeV.  $N_{\gamma}$  was observed to be generally between 1 and 5 for  $E^*$  between 3 and 10 MeV for the (p,p'), (d,p), (d, $\alpha$ ), (p, $\alpha$ ), and ( $\alpha$ , $\alpha'$ ) reactions investigated, and somewhat higher for the ( $\alpha$ ,p) reactions.  $N_{\gamma}$  increases with  $E^*$ , and is larger for higher angular momentum transfer reactions although it is not as much larger as angular momentum transfer considerations alone would suggest. (This can be explained as a level-d. effect for compound-nucleus reactions- the average angular momentum of the **states excited** by the reactions is severely limited by the level-d. spin-cutoff parameter). The dependence of  $N_{\gamma}$  on the average excess angular momentum of the residual nuclei was investigated by using calculated spin distributions for the residual nuclei. This dependence is similar to that for n capture  $N_{\gamma}$ 's- $N_{\gamma}$  increases with excess angular momenta for excess angular momenta of >2 or 3 units.

IT Gamma ray

(from nuclear reactions, multiplicity of)

IT 14314-35-3, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(deuteron bombardment of, multiplicity of  $\gamma$ -rays from)

RN 14314-35-3 HCAPLUS

CN Tin, isotope of mass 119

$^{119}\text{Sn}$

IT 10198-40-0, properties

RL: RCT (Reactant); RACT (Reactant or reagent)

(gamma rays from, from  $\alpha$ -particle bombardment of iron-57, multiplicity of)

RN 10198-40-0 HCAPLUS

CN Cobalt, isotope of mass 60

$^{60}\text{Co}$

**L128 ANSWER** 32 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1971:428388 HCAPLUS  
 DN 75:28388  
 OREF 75:4471a,4474a  
 TI Interpolative formulas for average nuclear level spacing and total radiation width  
 AU Musgrove, Anthony R.  
 CS Res. Establ., Aust. At. Energ. comm., Lucas Heights, Australia  
 SO Aust. At. Energy Comm., AAEC/E [Rep.] (1970), AAEC/E211, 6 pp.  
 CODEN: AATCAZ  
 DT Report  
 LA English  
 CC 75 (Nuclear Phenomena)  
 AB The free gas model formula for nuclear level ds. was used to interpolate unknown level spacings. The level d. parameter was fitted semi-empirically to give good fits to all exptl. measured level spacings. A formula for total radiation width is given which relates this quantity to the mass number, the average level spacing, and the excitation energy. Compound nuclei which have a long-**lived** isomeric state were found to have, on average, a radiation width 25 less than for non-isomeric compound nuclei. No correlation between radiation widths and n strength function was observed within the limits of accuracy of the exptl. data, contrary to some previous suggestions. Tables of calculated and measured resonance parameters are given for target nuclei in the range  $60 < A < 203$ . The calculated 30 keV capture cross section is compared with measured and semiempirical data (Allen et al., 1970) and with the calculated values of Benzi and Reffo (1969).  
 IT Nuclear energy levels  
     (spacing and width of, calcn. of)  
 IT **Gamma** rays  
     (transitions of, calcn. of)  
 IT **14191-71-0**, reactions  
     RL: RCT (Reactant); RACT (Reactant or reagent)  
     (neutron capture cross sections and radiation width of, calcn. of)  
 RN 14191-71-0 HCAPLUS  
 CN Indium, isotope of mass 115 (CA INDEX NAME)  
  
 115In  
 IT **10198-40-0**, properties  
     RL: PRP (Properties)  
     (nuclear energy level spacing and radiation width of, calcn. of)  
 RN 10198-40-0 HCAPLUS  
 CN Cobalt, isotope of mass 60 (CA INDEX NAME)  
  
 60Co

**L128 ANSWER** 34 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1970:95576 HCAPLUS  
 DN 72:95576  
 OREF 72:17365a,17368a  
 TI Photon and electron **excitation** of nuclear isomers  
 AU Booth, Edward C.  
 CS Boston Univ., Boston, MA, USA  
 SO U.S. Clearinghouse Fed. Sci. Tech. Inform., AD (1969),  
 AD-697153, 7 pp. Avail.: CFSTI  
 From: U. S. Govt. Res. Develop. Rep. 1970, 70(2), 167  
 CODEN: XCCIAV  
 DT Report  
 LA English  
 CC 75 (Nuclear Phenomena)  
 AB Isomeric states of stable nuclei have been **excited** by photons and e via **short-lived** higher states. The isomer **excitation** expts. were undertaken in order to determine the relation between the total inelastic e scattering cross section and the radiative width, and to use the **isomer excitation function to determine the energies widths, and angular momenta of accessible states**. The following isotopes were studied: 87Sr, 103Rh, 111Cd, 113In, 115In, 117Sn, 135Ba, 137Ba, 195Pt, and 199Hg.  
 IT **Gamma rays**  
 (bombardment by, nuclear isomeric **excitation** in)  
 IT Nuclear energy levels  
 (**excitation** of, by electrons and **gamma** rays)  
 IT 13981-59-4, properties 14191-71-0, properties  
 14191-88-9, properties 14336-64-2, properties  
 RL: PRP (Properties)  
 (nuclear energy levels of, **excited** by electrons and  
**gamma** rays)  
 RN 13981-59-4 HCAPLUS  
 CN Tin, isotope of mass 117 (CA INDEX NAME)  
 117Sn  
 RN 14191-71-0 HCAPLUS  
 CN Indium, isotope of mass 115 (CA INDEX NAME)  
 115In  
 RN 14191-88-9 HCAPLUS  
 CN Platinum, isotope of mass 195 (CA INDEX NAME)  
 195Pt  
 RN 14336-64-2 HCAPLUS  
 CN Cadmium, isotope of mass 111 (CA INDEX NAME)  
 111Cd

**L128 ANSWER** 35 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1970:17446 HCAPLUS

DN 72:17446

OREF 72:3187a,3190a

TI Nuclear photoactivation of selenium-77, silver-107, silver-109, cadmium-111, indium-115, and mercury-199

AU Boivin, Michel; Cauchois, Yvette; Heno, Yvonne

CS Lab. Chim. Phys., Fac. Sci. Paris, Paris, Fr.

SO Nuclear Physics A (1969), 137(3), 520-30

CODEN: NUPABL; ISSN: 0375-9474

DT Journal

LA English

CC 75 (Nuclear Phenomena)

AB The photoactivation of various isotopes is studied by using the **x-ray** bremsstrahlung produced by a 2-MeV e accelerator. The photoactivation cross section was calculated for each of the **observed nuclear levels**. Some of these levels were observed for the 1st time. Their probable characteristics are proposed or confirmed. In certain cases it was possible to determine partial widths and the **mean life of the level**.

**X-rays**

(bombardment by, of selenium-77 and silver isotopes)

IT 14191-71-0, properties 14336-64-2, properties

RL: PRP (Properties)

(nuclear energy levels of, from **x-ray** bombardment)

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

115In

RN 14336-64-2 HCAPLUS

CN Cadmium, isotope of mass 111 (CA INDEX NAME)

111Cd

**L128 ANSWER** 36 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1969:466270 HCAPLUS

DN 71:66270

OREF 71:12237a,12240a

TI **Photoactivation with a cobalt-60 source**

AU Law, J.; Iddings, F. A.

CS Longwood Coll., Farmville, VA, USA

SO Journal of Radioanalytical Chemistry (1969), 3(1-2), 53-63

CODEN: JRACBN; ISSN: 0022-4081

DT Journal

LA English

CC 75 (Nuclear Phenomena)

AB Photoactivation of 13 elements was studied by using 5000-Ci. and 30,000-Ci. sources. Reported data from the study are presented in tables including exptl. conditions, observed activity, sensitivity achieved, . **gamma**. energies, and **half-life** observed, and reaction cross sections. Comparisons are made between this research and earlier reports.

IT **14191-71-0**, analysis **14336-64-2**, analysis

RL: ANT (Analyte); ANST (Analytical study)  
(detection of, by **gamma**-ray activation)

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

115In

RN 14336-64-2 HCAPLUS

CN Cadmium, isotope of mass 111 (CA INDEX NAME)

111Cd

**L128 ANSWER** 37 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1967:459835 HCAPLUS  
 DN 67:59835  
 OREF 67:11238e  
 TI **Excitation** of metastable levels in nuclei by photoactivation on indium-gallium radiation loop  
 AU Abrams, I.; Pelekis, L.; Taure, I.  
 SO Neitronoakt. Anal. (1966), 135-40  
 CODEN: 16GJAI  
 DT Conference  
 LA Russian  
 CC 75 (Nuclear Phenomena)  
 AB The isomers were obtained in an emitter of a radiation loop sphere-shaped and 15 cm. in diameter with a channel passing through it, 5 cm. in diameter. An eutectic alloy (Sn 13, In 15, Ga 62%) was used as a circulating substance. The flux of  $\gamma$ -quanta in the middle of the sphere was  $4 + 10^{12}$   $\gamma$ -quanta/cm.2-sec. The following data were obtained: Isomer 111Cd:  $T_{1/2} = 47 \pm 1$  min.,  $E\gamma = 152 \pm 4$  and  $246 \pm 4$  kev., cross section  $(6.9 \pm 1.4) + 10^{-6}$  mb.; 115In:  $T_{1/2} = 4.23 \pm 0.08$  hr.,  $E\gamma = 335 \pm 4$  kev., cross section  $(1.5 \pm 0.3) + 10^{-4}$  mb.  
 IT **Gamma rays**  
     (facilities, of gallium-indium alloy loop, **excitation** of metastable levels in)  
 IT Nuclear energy levels  
     (metastable, photoactivation in gallium-indium alloy radiation loop)  
 IT 14191-71-0, properties 14336-64-2, properties  
 RL: PRP (Properties)  
     (**gamma**-rays from and **half-life** of metastable)  
 RN 14191-71-0 HCAPLUS  
 CN Indium, isotope of mass 115 (CA INDEX NAME)

115In

RN 14336-64-2 HCAPLUS  
 CN Cadmium, isotope of mass 111 (CA INDEX NAME)

111Cd

**L128 ANSWER** 38 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1964:408987 HCAPLUS

DN 61:8987

OREF 61:1436e-g

**TI Production of nuclear isomers. I. Production of nuclear isomers with bremsstrahlung**

AU Kaminishi, T.; Kojima, C.

SO Nagoya Kogyo Gijutsu Shikensho Hokoku (1962), 11(5), 284-95

From: Phys. Abstr. 65(778), Abstr. No. 18271(1962).

CODEN: NKGSAR; ISSN: 0027-7614

DT Journal

LA Unavailable

CC 12 (Nuclear Phenomena)

AB A number of stable elements were irradiated with bremsstrahlung from a 6-m.e.v. linear e accelerator to produce directly each nuclear isomer which has been confirmed by other nuclear reactions. After irradiation the sample was moved to the counting system to measure the radioactivity produced and the isomeric nuclide was determined from the half-life and  $\gamma$ -ray energy of its decay. The production of the following isomers was confirmed: 77Sem, 79Brm, 87Srm, 89Ym, 103Rhm, 107Agm + 109Agm, 111Cdm, 113Inm + 115Inm, 117Sn, 135Bam + 137Bam, 167Erm, 179Hfm, 183Wm, 191Irm, 195Pt, 197Aum, and 199Hgm. Radioactive substances with **halflives** of 150 sec. in the V sample and 2.0 sec. in the W sample were produced, but not every radioactive nuclide could be determined because no information was available. Not all isomers of even-even nuclei were observed, although some **half-lives** and . **gamma**.-ray energies could be easily measured by this method if these isomers were produced.

IT **Gamma rays**

(nuclear isomer production by)

IT **14191-88-9P**, Platinum, isotope of mass 195    **14265-76-0P**, Hafnium, isotope of mass 179

RL: PREP (Preparation)

(formation of metastable, by bremsstrahlung irradiation)

RN 14191-88-9 HCAPLUS

CN Platinum, isotope of mass 195 (CA INDEX NAME)

195Pt

RN 14265-76-0 HCAPLUS

CN Hafnium, isotope of mass 179 (CA INDEX NAME)

179Hf

IT **14191-71-0P**, Indium, isotope of mass 115

RL: PREP (Preparation)

(formation of metastable, by bremsstrahlung radiation)

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

115In

IT **13981-59-4P**, Tin, isotope of mass 117

RL: PREP (Preparation)

(production of metastable, by bremsstrahlung)

RN 13981-59-4 HCAPLUS

CN Tin, isotope of mass 117 (CA INDEX NAME)

117Sn

**L128 ANSWER** 39 OF 42 HCPLUS COPYRIGHT ACS on STN

AN 1953:54226 HCPLUS  
 DN 47:54226  
 OREF 47:9170f-g  
 TI Angular correlation of nickel60  
 AU Wiedling, T.  
 CS Univ. Stockholm  
 SO Arkiv foer Fysik (1953), 7, 69-71  
 CODEN: AFYSA7; ISSN: 0365-2440  
 DT Journal  
 LA English  
 CC 3A (Nuclear Phenomena)  
 AB A Co60 source in the form of CoCl<sub>2</sub> gave an anisotropy,  $(W(\pi)/W(\pi/2)) - 1$ , of  $0.147 \pm 0.007$ .  
 IT Atomic nuclei  
     (deuteron-bombarded Zn)  
 IT Gamma rays  
     (from cadmium-111, angular correlations of)  
 IT 10198-40-0, Cobalt, isotope of mass 60  
     (angular correlation of radiation from)  
 RN 10198-40-0 HCPLUS  
 CN Cobalt, isotope of mass 60 (CA INDEX NAME)  
  
 $^{60}\text{Co}$   
 IT 14336-64-2, Cadmium, isotope of mass 111  
     (angular correlation of  $\gamma$ - $\gamma$ -  
     cascade from)  
 RN 14336-64-2 HCPLUS  
 CN Cadmium, isotope of mass 111 (CA INDEX NAME)

111cd

L128 ANSWER 41 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1950:55568 HCAPLUS

DN 44:55568

OREF 44:10539a-c

TI **Radioactivity from enriched isotopes of cadmium**

AU Cork, J. M.; Rutledge, W. C.; Stoddard, A. E.; Branyan, C. E.; LeBlanc, J. M.

CS Univ. of Michigan, Ann Arbor

SO Physical Review (1950), 79, 938-9

CODEN: PHRVAO; ISSN: 0031-899X

DT Journal

LA Unavailable

CC 3A (Nuclear Phenomena)

AB By irradiating in the pile specimens of Cd enriched in Cd108 and in Cd114, it is possible to interpret more positively the observed radioactivities and to evaluate the  $\gamma$ -rays associated with each. Cd109 produced by neutron capture in Cd108 decays with a half life of .apprx.250 days by K-capture to Ag109, in which a strong .gamma .-ray of 87.5 e.kv. is emitted. Cd115 is isomeric with half lives of 54 hrs. and 42.6 days, both isomers decaying to In115 by  $\beta$ -emission. The 54-hr. activity yields  $\gamma$ -rays of energies 335.5, 343.7, 348.9, 369.3, 423.7, 451.9, 525.4, 559.1, and 713.1 e.kv. These  $\gamma$ -energies fit well a proposed nuclear level scheme in In115.

IT Gamma rays

(from cadmium isotopes)

IT Atomic nuclei

(isomerism of Sn)

IT **14191-71-0**, Indium, isotope of mass 115

(from decay of Cd115)

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

<sup>115</sup>In

IT **13981-59-4**, Tin, isotope of mass 117 **14314-35-3**,

Tin, isotope of mass 119

(isomeric)

RN 13981-59-4 HCAPLUS

CN Tin, isotope of mass 117 (CA INDEX NAME)

<sup>117</sup>Sn

RN 14314-35-3 HCAPLUS

CN Tin, isotope of mass 119 (CA INDEX NAME)

<sup>119</sup>Sn

**L128 ANSWER** 42 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1950:9465 HCAPLUS  
 DN 44:9465  
 OREF 44:1814b-e  
 TI Coincidence arrangement for scintillation counter  
 AU Meyer, K. P.; Baldinger, E.; Hahn, B.; Huber, P.  
 CS Univ. Basel, Switz.  
 SO Helvetica Physica Acta (1949), 22, 420-4  
 CODEN: HPACAK; ISSN: 0018-0238  
 DT Journal  
 LA Unavailable  
 CC 3A (Nuclear Phenomena)  
 AB The apparatus consists of an anthracene crystal, two photomultiplier tubes, and a coincidence-mixing circuit containing Ge diodes. The background is reduced by a factor of 500 compared with that for a single photomultiplier counting at the rate of 10,000 per sec. The discrimination curve of the circuit gives a plateau so that the number of counts registered depends only on the source, crystal, and their relative positions and is independent of the photomultipliers and the amplification. In order to count coincident events, a second crystal and a third photomultiplier tube are added to the circuit. With this arrangement and a **Co60** source of 0.01 millicurie, measurements were made on the directional correlation of the two ;. **gamma.-rays emitted in cascade.** If  $K'$  and  $K''$  are the number of true coincidences per min., at  $180^\circ$  and  $90^\circ$ , resp.,  $\eta = (K' - K'')/K'' = 15 \pm 2\%$ . The background was 1 count per min., the space angle was 0.01, and the number of accidental and true coincidences per min. were 6 and 60, resp.  
 IT **Gamma** rays  
     (from cobalt-60)  
 IT **Gamma** rays  
     (from indium-115)  
 IT **14191-71-0**, Indium, isotope of mass 115  
     ( $\beta$ - and  $\gamma$ -rays from)  
 RN 14191-71-0 HCAPLUS  
 CN Indium, isotope of mass 115 (CA INDEX NAME)

$^{115}\text{In}$

IT **10198-40-0**, Cobalt, isotope of mass 60  
     ( $\gamma$ -rays from)  
 RN 10198-40-0 HCAPLUS  
 CN Cobalt, isotope of mass 60 (CA INDEX NAME)

$^{60}\text{Co}$

**CAS/STN FILE 'REGISTRY' ENTERED AT 13:29:44 ON 04 DEC 2009**

L1 137 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?93NB?/CNS  
 L2 36 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?111CD?/CNS  
 L3 27 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?113CD?/CNS  
 L4 6 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?135CS?/CNS  
 L5 62 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?115IN?/CNS  
 L6 97 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?117SN?/CNS  
 L7 88 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?119SN?/CNS  
 L8 106 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?125TE?/CNS  
 L9 98 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?129XE?/CNS  
 L10 61 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?131XE?/CNS  
 L11 12 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?178HF?/CNS  
 L12 9 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?179HF?/CNS  
 L13 14 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?193IR?/CNS  
 L14 89 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?195PT?/CNS

FILE 'STNGUIDE' ENTERED AT 13:30:02 ON 04 DEC 2009

FILE 'REGISTRY' ENTERED AT 13:32:12 ON 04 DEC 2009

L15 835 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON (L1 OR L2 OR L3 OR  
     L4 OR L5 OR L6 OR L7 OR L8 OR L9 OR L10 OR L11 OR L12 OR L13 OR L14)  
 L16 773 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON (L1 OR L2 OR L3 OR  
     L4) OR (L6 OR L7 OR L8 OR L9 OR L10 OR L11 OR L12 OR L13 OR L14)  
 L17 62 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?115IN?/CNS

FILE 'STNGUIDE' ENTERED AT 13:32:13 ON 04 DEC 2009

FILE 'HCAPLUS' ENTERED AT 13:34:27 ON 04 DEC 2009

L18 625 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L4  
 L19 95 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L1  
 L20 1298 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L2  
 L21 982 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L3  
 L22 1817 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L5  
 L23 930 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L6  
 L24 2387 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L7  
 L25 1198 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L8  
 L26 2129 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L9  
 L27 1193 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L10  
 L28 752 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L11  
 L29 467 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L12  
 L30 524 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L13  
 L31 1296 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L14

FILE 'REGISTRY' ENTERED AT 13:34:56 ON 04 DEC 2009

L32 62 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON 60CO?/CNS

FILE 'HCAPLUS' ENTERED AT 13:34:57 ON 04 DEC 2009

L33 10405 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L32

FILE 'LCA' ENTERED AT 13:35:41 ON 04 DEC 2009

FILE 'HCAPLUS' ENTERED AT 13:36:18 ON 04 DEC 2009

L34 1 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON US20080078961/PN  
 L35 SEL PLU=ON L34 1- RN : 14 TERMS  
 L36 193 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L35  
 L37 1 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L34 AND L36

FILE 'STNGUIDE' ENTERED AT 13:36:36 ON 04 DEC 2009

FILE 'LCA' ENTERED AT 13:37:20 ON 04 DEC 2009

L38 1992 SEA FILE=LCA SPE=ON ABB=ON PLU=ON GAMMA  
 L39 96 SEA FILE=LCA SPE=ON ABB=ON PLU=ON CASCAD?  
 L40 0 SEA FILE=LCA SPE=ON ABB=ON PLU=ON RADIOCASCAD?  
 L41 677 SEA FILE=LCA SPE=ON ABB=ON PLU=ON DECAY?  
 L42 6 SEA FILE=LCA SPE=ON ABB=ON PLU=ON HALFLI?  
 L43 230 SEA FILE=LCA SPE=ON ABB=ON PLU=ON HALF LIFE  
 L44 57 SEA FILE=LCA SPE=ON ABB=ON PLU=ON HALF LIVES  
 L45 200 SEA FILE=LCA SPE=ON ABB=ON PLU=ON (LIFE OR LIVE##) (2A) (SHORT##### OR LONG##### OR LENGTH#####)

**CAS/STN (CONTINUED)**

## FILE 'LCA' (CONTINUED)

L46 0 SEA FILE=LCA SPE=ON ABB=ON PLU=ON LONGLIF?  
 L47 1 SEA FILE=LCA SPE=ON ABB=ON PLU=ON LONGLIV?  
 L48 0 SEA FILE=LCA SPE=ON ABB=ON PLU=ON SHORTLIV?  
 L49 0 SEA FILE=LCA SPE=ON ABB=ON PLU=ON SHORTLIF?  
 L50 27 SEA FILE=LCA SPE=ON ABB=ON PLU=ON DEEXCIT? OR DE EXCIT?  
 L51 1573 SEA FILE=LCA SPE=ON ABB=ON PLU=ON EXCIT?  
 L52 160 SEA FILE=LCA SPE=ON ABB=ON PLU=ON (MIX##### OR BLEND##### OR PAIR##### OR DOUBLE OR  
 TWO OR SECOND OR 2ND OR COUPL##### OR ANOTHER) (3A) (?ISOMER? OR ?NUCLIDE? OR NUCLEIDE? OR ?SPECIES?)  
 L53 47 SEA FILE=LCA SPE=ON ABB=ON PLU=ON (MIX##### OR BLEND#####  
 OR PAIR##### OR DOUBLE OR TWO OR SECOND OR 2ND OR COUPL##### OR ANOTHER) (3A) ?ISOTOP?  
 L54 24 SEA FILE=LCA SPE=ON ABB=ON PLU=ON ENTANGL#####

## FILE 'HCAPLUS' ENTERED AT 13:42:46 ON 04 DEC 2009

L55 73 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L22 AND L33  
 L56 115 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L18 OR L19 OR L20 OR L21) AND L33  
 L57 153 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L23 OR L24 OR L25 OR L26 OR L27 OR L28 OR L29  
 OR L30 OR L31) AND L33  
 L58 236 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L23 OR L24 OR L25 OR  
 L26 OR L27 OR L28 OR L29 OR L30 OR L31) AND L22  
 L59 230 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L18 OR L19 OR L20 OR L21) AND L22

## FILE 'STNGUIDE' ENTERED AT 13:42:48 ON 04 DEC 2009

## FILE 'HCAPLUS' ENTERED AT 13:45:05 ON 04 DEC 2009

L60 142 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L58 OR L59) AND  
 (GAMMA OR X(2A) (RAY OR RADIATION OR PHOTON#####))  
 L61 73 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L22 AND L33  
 L62 170 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L56 OR L57) NOT L55  
 L63 368 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L60 OR L61 OR L62)  
 L64 197 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L38  
 L65 9 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L39  
 L66 0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L40  
 L67 65 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L41  
 L68 2 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L42  
 L69 20 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L43  
 L70 13 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L44  
 L71 31 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L45  
 L72 0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L46  
 L73 0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L47  
 L74 0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L48  
 L75 0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L49  
 L76 1 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L50  
 L77 59 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L51  
 L78 6 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L52  
 L79 3 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L53  
 L80 1 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L54  
 L81 SEL PLU=ON L34 1- IPC ECLA FTERM NCL : 5 TERMS  
 L82 6074 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L81  
 L83 1 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L82

## FILE 'STNGUIDE' ENTERED AT 13:48:04 ON 04 DEC 2009

## FILE 'HCAPLUS' ENTERED AT 13:50:07 ON 04 DEC 2009

L84 19 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND X(2A) (RAY OR  
 RADIATION OR PHOTON#####)  
 L85 45 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L64 AND L67  
 L86 53 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L64 AND L77  
 L87 17 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L67 AND L77

## FILE 'LCA' ENTERED AT 13:50:49 ON 04 DEC 2009

## FILE 'HCAPLUS' ENTERED AT 13:52:25 ON 04 DEC 2009

L88 135 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L65 OR (L68 OR L69 OR  
 L70 OR L71 OR L72 OR L73 OR L74 OR L75 OR L76) OR (L78 OR L79  
 OR L80) OR L83 OR (L84 OR L85 OR L86 OR L87)  
 L89 135 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L88 NOT L34  
 L90 53 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L89 AND 1940-1980/PY

**CAS/STN (CONTINUED)**

## FILE 'HCAPLUS' (CONTINUED)

L91           74 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L89 AND 1981-2005/PY  
 L92           0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L89 AND 1981-2004/PRY  
 L93           0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L89 AND 2005/PRY AND (WO OR US)/PRC  
 L94           127 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L90 OR L91  
 L95           0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND ENTANGL#####(6A)GAMMA  
 L96           480 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON ENTANGL##### AND GAMMA  
 L97           0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND L96  
 L98           0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND ENTANGL?  
 L99           3 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND (MIX##### OR BLEND##### OR PAIR##### OR  
 DOUBLE OR TWO OR SECOND OR 2ND OR COUPL##### OR ANOTHER) (3A)?ISOTOP?  
 L100          6 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND (MIX##### OR BLEND##### OR PAIR##### OR  
 DOUBLE OR TWO OR SECOND OR 2ND OR COUPL##### OR ANOTHER) (3A)?ISOMER? OR ?NUCLIDE? OR NUCLEIDE? OR  
 ?SPECIES?)  
 L101          41 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND L5(L)GAMMA  
 L102          0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND L5(L)CASCAD?  
 L103          7 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L5(L)CASCAD?  
 L104          0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND L5(L)?CO60?  
 L105          2 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND L5(L)?60CO?  
 L106          0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND L5(L)CO  
 L107          0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND L5(L)COBALT  
 L108          4 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L5(L)COBALT  
 L109          5 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L5(L)CO  
 L110          73 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L22 AND L33  
 L111          30 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L110 AND GAMMA

FILE 'LCA' ENTERED AT 13:58:08 ON 04 DEC 2009

FILE 'HCAPLUS' ENTERED AT 13:59:44 ON 04 DEC 2009

L112          121 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L65 OR L68 OR L70 OR  
               L76 OR (L78 OR L79 OR L80) OR L84 OR L87 OR (L99 OR L100 OR  
               L101 OR L102 OR L103 OR L104 OR L105 OR L106 OR L107 OR L108 OR L109) OR L111  
 L113          1 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND L36  
 L114          1 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L113 NOT L34  
 L115          122 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L112 OR L113 OR L114) NOT L34  
 L116          64 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L115 AND 1980-2005/PY  
 L117          50 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L115 AND 1940-1979/PY  
 L118          114 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L116 OR L117)  
 L119          1 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L118 AND DEEXCIT?  
 L120          0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L118 AND DE EXCIT?  
 L121          10 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L118 AND EXCIT#####(2A)STATE  
 L122          20 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L118 AND LIFE  
 L123          20 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L118 AND LIVE#  
 L124          2 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L118 AND HALFLI?  
 L125          0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L118 AND SHORTLI?  
 L126          0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L118 AND LONGLI?  
 L127          6 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L118 AND (GAMMA OR X)(5A)CASCAD?  
 L128          42 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L114 OR (L119 OR L120  
               OR L121 OR L122 OR L123 OR L124 OR L125 OR L126 OR L127)  
               D ALL HITSTR TOT

**EAST Search History for a search on 12-4-2009:**

12	Patent	57 "5674127" or tu adj "2064697"	US-AGRUS; USPAT; EPO; PTO; DERVENT; IBM_TDB	OR	ON	2009/12/04
13	Patent	29 U3 and (gamma or (S or S)) with (second or two or composition or composite or mix\$3 or blend\$3 or specie\$5 or :radioisomer\$5 or nucleo\$5 or species or isotop\$5 or radionucl\$5 or species or isotop\$5 or radionucl\$5 or species or isotop\$5 or radionucl\$5 or species or isotop\$5 or rad...)	US-AGRUS; USPAT; EPO; PTO; DERVENT; IBM_TDB	OR	ON	2009/12/04
15	Patent	26 U3 and (level or entangl\$7 or energ\$7 or exsol\$7 or deexch\$7) with (second or two or composition or composite or mix\$3 or blend\$3 or :isomer\$5 or :radioisomer\$5 or nucleo\$5 or nucleo\$5 or radionucl\$5 or species or isotop\$5 or radionucl\$5 or species or isotop\$5 or rad...)	US-AGRUS; USPAT; EPO; PTO; DERVENT; IBM_TDB	OR	ON	2009/12/04
16	Patent	42 U3 and (level or entangl\$7 or energ\$7 or exsol\$7 or deexch\$7 or second or two or composition or composite or mix\$3 or blend\$3 or :isomer\$5 or :radioisomer\$5 or nucleo\$5 or nucleo\$5 or radionucl\$5 or species or isotop\$5 or rad...)	US-AGRUS; USPAT; EPO; PTO; DERVENT; IBM_TDB	OR	ON	2009/12/04
17	Patent	26 U4 and U5 and U6	US-AGRUS; USPAT; EPO; PTO; DERVENT; IBM_TDB	OR	ON	2009/12/04
18	Patent	2 "5674127" or tu adj "2064697"	US-AGRUS; USPAT; EPO; PTO; DERVENT; IBM_TDB	OR	ON	2009/12/04
19	Patent	12 U7 and (frosts\$ or (w\$4))	US-AGRUS; USPAT; EPO; PTO; DERVENT; IBM_TDB	OR	ON	2009/12/04
20	Patent	23 U3 and (level or entangl\$7 or energ\$7 or exsol\$7 or deexch\$7 or second or two or composition or composite or mix\$3 or blend\$3 or :isomer\$5 or :radioisomer\$5 or nucleo\$5 or nucleo\$5 or radionucl\$5 or species or isotop\$5 or rad...)	US-AGRUS; USPAT; EPO; PTO; DERVENT; IBM_TDB	OR	ON	2009/12/04
21	Patent	U3 and (S4 or U5 or U6)	US-AGRUS; USPAT; EPO; PTO; DERVENT; IBM_TDB	OR	ON	2009/12/04

EAST Search 11-30-09:

**EAST Search History (Prior Art)**

Ref #	Hits	Search Query	DBs	Default Operator	Plurals	Time Stamp
L1	0	"2008315210"	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 08:08
L2	1	"20080315210"	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 08:08
L3	1	"20030078961"	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 08:29
L4	1	2005-716517.NRAN.	DERWENT	OR	OFF	2009/11/30 08:33
L7	0	(mix\$6 or blend\$5) near5 two adj1 (isotopes or radioisotopes or isomers or radioisomers) and (half adj lrf\$2 or half adj lrv\$2)	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:37
L8	0	(mix\$6 or blend\$5) near5 two adj1 (isotopes or radioisotopes or isomers or radioisomers) and decay\$5	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:41
L9	21	(mix\$6 or blend\$5) near5 (isotopes or radioisotopes or isomers or radioisomers) and decay\$5	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:41
L10	0	(mix\$6 or blend\$5) near5 (isotopes or radioisotopes or isomers or radioisomers) and halflif\$4	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:41
L11	0	(mix\$6 or blend\$5) near5 (isotopes or radioisotopes or isomers or radioisomers) and halfliv\$4	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:41
L12	4	(mix\$6 or blend\$5) near5 (isotopes or radioisotopes or isomers or radioisomers) and half adj lrv\$4	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:42
L13	22	(mix\$6 or blend\$5) near5 (isotopes or radioisotopes or isomers or radioisomers) and half adj lrf\$4	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:42
L14	42	L9 L12 L13	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:42
L15	1	L14 and shift\$6	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:43
L16	1	L14 and curv\$5	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:43

**EAST Search History (Prior Art)**

L17	2	L14 and g21k\$	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:44
L18	3	L14 and k09-e\$	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:44
L19	5	L17 L18	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:44
L20	29639	g21k\$	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:50
L21	4010	k09-e\$	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:51
L22	33573	L20 L21	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:51
L23	3	L22 and decay\$5 near5 shorter	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:51
L24	12	L22 and decay\$5 near5 short\$8	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:51
L25	7	L22 and decay\$5 near5 long\$5	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:51
L26	0	L22 and decay\$5 near5 length\$5	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:51
L27	1	L22 and decay\$5 near5 curv\$6	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:52
L28	19	L23 L24 L25 L27	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:52
L29	5	L28 and gamma	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:22
L30	152	gamma near7 decay\$7	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:24
L31	0	L30 and (mix\$6 or blend\$6) near4 two	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:25

## EAST Search History (Prior Art)

L32	1	L30 and (mix\$6 or blend\$6 or two or second) near2 (isomer\$5 or radioisomer\$5 or radioisotop\$6 or isotop\$6 or nuclid\$5 or nucleid\$6 or radio or radionuc\$9)	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:26
L33	0	L30 and composite near3 half	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:27
L34	18	L30 and lives	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:28
L35	18	L30 and lifes	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:28
L36	14	L30 and half	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:28
L37	19	L34-L35 L36	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:28
L38	0	L37 and composite	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:29
L39	0	L37 and mix\$5	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:29
L40	0	L37 and blend\$5	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:29
L41	6	L37 and two	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:29
L42	4	L37 and second	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:29
L43	10	L41 L42	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:29
L44	1178	(vary\$6 or vari\$5 or chang\$6 or increas\$5 or decreas\$6) near6 half adj life	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:32
L45	31	(vary\$6 or vari\$5 or chang\$6 or increas\$5 or decreas\$6) near6 half adj life near6 (composite or blend\$6 or mix\$6 or effective)	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:32

## EAST Search History (Prior Art)

L46	31	(vary\$6 or vari\$5 or chang\$6 or increas\$5 or decreas\$6) near6 half adj lives near6 (composite or blend\$6 or mix\$6 or effective)	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:32
L47	0	(vary\$6 or vari\$5 or chang\$6 or increas\$5 or decreas\$6) near6 halflives near6 (composite or blend\$6 or mix\$6 or effective)	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:33
L48	0	(vary\$6 or vari\$5 or chang\$6 or increas\$5 or decreas\$6) near6 halflife near6 (composite or blend\$6 or mix\$6 or effective)	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:33
L49	31	L45-L46	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:33
L50	0	L49 and decay\$5 near4 curv\$5	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:33
L51	0	L49 and decay\$5	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:33
L52	0	L49 and curv\$5	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:33
L53	0	L49 and gamma	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:34
L54	0	L49 and skew\$6 near6 effective adj half adj life	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:34
L55	0	skew\$6 near6 effective adj half adj life	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:34
L56	2	graph\$5 near6 effective adj half adj life	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:34
L57	0	curv\$6 near6 effective adj half adj life	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:34
L58	0	skew\$6 near6 effective adj half adj life	US-PGPUB; USPAT; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:35

**EAST Search History (Prior Art)**

L59	1	curv\$5 near6 effective adj half adj life	US-PGPUB; USPAT; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:35
L60	4	graph\$6 near6 effective adj half adj life	US-PGPUB; USPAT; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:36
L61	0	fig near6 effective adj half adj life	US-PGPUB; USPAT; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:36
L62	0	figure near6 effective adj half adj life	US-PGPUB; USPAT; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:36
L63	5	L56 L59 L60	US-PGPUB; USPAT; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:36
L64	6862	(long\$6 near7 short\$5) near7 (life or lived or live or halflife or halflive\$1)	US-PGPUB; USPAT; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:44
L65	10	(long\$6 near7 short\$5) near7 (life or lived or live or halflife or halflive\$1) near7 (isptop\$5 or radioisotop\$6 or nuclide or radionu\$8 or nucleide or isomer\$6 or radiolsomer\$6) near7 (two or second)	US-PGPUB; USPAT; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:46
L67	0	hybrid near4 radiation near4 source near12 (decays\$6 or long\$5 or short\$6 or half or halfl\$6) near12 gamma	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:54
L68	0	hybrid near4 (radiation or source) near12 (decay\$6 or long\$5 or short\$6 or half or halfl\$6) near12 gamma	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:55

**EAST Search History (Prior Art)**

L69	47	hybrid near4 radiation near4 source	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:55
L70	0	L69 and g21k\$6	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:57
L71	2	(nix\$7 or blend\$6) near4 halflife	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:57
L72	36	(nix\$7 or blend\$6) near4 half adj life	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:57
L73	36	L71 L72	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:58
L74	10	L64 and g21k\$	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:00
L75	56	"5674177"	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:06
L76	25	L75 and (long\$5 or short\$6) near5 (life or lives or halfli\$6 or lived or decay\$6)	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:07

## EAST Search History (Prior Art)

L77	10	L75 and (relat\$4 or relation\$7 or relativ\$5 or compar\$4 or comparat\$6) near5 (life or lives or halfli\$6 or lived or decay\$6)	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:09
L78	10	L76 and L77	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:09
L79	1008	(curv\$5 or graph\$5 or fig or figure) near4 (half adj life)	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:12
L80	0	(curv\$5 or graph\$5 or fig or figure) near7 ((half adj life) or halfli\$7) near7 (hybrid or composite or mix\$3 or mixture or blend\$5 or two) near7 (radio or radioactiv\$6 or radioisotop\$6 or nuclide or nucleide or radionu\$7 or radioisomer\$6 or isomer\$5)	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:15
L81	89	(curv\$5 or graph\$5 or fig or figure) same ((half adj life) or halfli\$7) same (hybrid or composite or mix\$3 or mixture or blend\$5 or two) near7 (radio or radioactiv\$6 or radioisotop\$6 or nuclide or nucleide or radionu\$7 or radioisomer\$6 or isomer\$5)	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:16
L82	20	L81 and shorter	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:16
L83	32	L81 and longer	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:17

**EAST Search History (Prior Art)**

L86	43	L81 and gamma near\$5 (EMIT\$6 OR EMISS\$5 or decay\$5)	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:18
L87	8	L81 and L82 and L83 and L84	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:19
L88	0	"3017514" and halflives	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	OFF	2009/11/30 11:21
L89	0	"3017514" and halflives	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	OFF	2009/11/30 11:21
L90	0	"3017514" and half adj lifies	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	OFF	2009/11/30 11:21
L91	1	"3017514" and half adj lifies	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	OFF	2009/11/30 11:21
L92	3	"3017514" and (composite or mix\$3 or mixture or blend\$6 or composition or compos\$4 or compris\$6) near3 two	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:23





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Art Unit \_\_\_\_\_

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Types     Patent(s)                       Non-Patent Literature

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